



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁵ : A61K 31/245, 31/195, 31/325 C07C 45/00	A1	(11) International Publication Number: WO 92/20335 (43) International Publication Date: 26 November 1992 (26.11.92)						
(21) International Application Number: PCT/US92/04229 (22) International Filing Date: 19 May 1992 (19.05.92) (30) Priority data: <table border="0"> <tr> <td>702,947</td> <td>20 May 1991 (20.05.91)</td> <td>US</td> </tr> <tr> <td>885,721</td> <td>18 May 1992 (18.05.92)</td> <td>US</td> </tr> </table> (71) Applicant: CENTER FOR INNOVATIVE TECHNOLOGY [US/US]; 2214 Rock Hill Road, Suite 600, Herndon, VA 22070 (US). (72) Inventors: ABRAHAM, Donald, J. ; 3511 Buckhead Road, Midlothian, VA 23113 (US). MAHRAN, Mona ; University of Alexandria, School of Pharmacy Medicinal Chemistry Department , Alexandria, 21215 (EG). MEHANNA, Ahmed ; 17 Bedford Road, Winchester, MA 01890 (US). RANDAD, Ramnarayan ; 7526 Exmouth Drive, Richmond, VA 23225 (US).		702,947	20 May 1991 (20.05.91)	US	885,721	18 May 1992 (18.05.92)	US	(74) Agent: WHITHAM, Michael, E.; 11800 Sunrise Valley Dr., Suite 220, Reston, VA 22091 (US). (81) Designated States: AT (European patent), BE (European patent), CA, CH (European patent), DE (European patent), DK (European patent), ES (European patent), FR (European patent), GB (European patent), GR (European patent), IT (European patent), JP, LU (European patent), MC (European patent), NL (European patent), SE (European patent). Published <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
702,947	20 May 1991 (20.05.91)	US						
885,721	18 May 1992 (18.05.92)	US						
(54) Title: USING ALLOSTERIC HEMOGLOBIN MODIFIERS TO DECREASE OXYGEN AFFINITY IN BLOOD (57) Abstract <p>A family of compounds has been found to be useful for right-shifting hemoglobin towards a low oxygen affinity state. The compounds are capable of acting on hemoglobin in whole blood. In addition, the compounds can maintain the oxygen affinity in blood during storage and can restore the oxygen affinity of outdated blood.</p>								

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	FI	Finland	ML	Mali
AU	Australia	FR	France	MN	Mongolia
BB	Barbados	GA	Gabon	MR	Mauritania
BE	Belgium	GB	United Kingdom	MW	Malawi
BF	Burkina Faso	GN	Guinea	NL	Netherlands
BG	Bulgaria	GR	Greece	NO	Norway
BJ	Benin	HU	Hungary	PL	Poland
BR	Brazil	IE	Ireland	RO	Romania
CA	Canada	IT	Italy	RU	Russian Federation
CF	Central African Republic	JP	Japan	SD	Sudan
CG	Congo	KP	Democratic People's Republic of Korea	SE	Sweden
CH	Switzerland	KR	Republic of Korea	SN	Senegal
CI	Côte d'Ivoire	LI	Liechtenstein	SU	Soviet Union
CM	Cameroon	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
DE	Germany	MC	Monaco	US	United States of America
DK	Denmark	MG	Madagascar		
ES	Spain				

USING ALLOSTERIC HEMOGLOBIN MODIFIERS TO
DECREASE OXYGEN AFFINITY IN BLOOD

CROSS-REFERENCE TO RELATED APPLICATIONS

5

This patent application is a continuation-in-part application of the co-pending U.S. Patent Application entitled "ALLOSTERIC HEMOGLOBIN MODIFIERS USEFUL FOR DECREASING OXYGEN AFFINITY AND PRESERVING OXYGEN CAPABILITY OF STORED BLOOD" having
10 Serial No. 07/702,947 which was filed May 20, 1991, and which issued as U.S. Patent _____. That patent application was itself a continuation-in-part application of now U.S. Patent 5,049,695 which was
15 filed on February 12, 1990. The subject matter of this application is also a continuation-in-part of the co-pending U.S. Patent Application entitled "ALLOSTERIC HEMOGLOBIN MODIFIER COMPOUNDS" having
20 Serial No. 07/722,382 which was filed June 26, 1991, and which itself is a continuation of the U.S. Patent Application having Serial No. 07/623,346 which was filed December 7, 1990. The text of all three of the above-identified patent applications and U.S. Patent is herein incorporated by reference.

25

DESCRIPTION

BACKGROUND OF THE INVENTION

30

Field of the Invention

The present invention generally relates to using a family of compounds to adjust the allosteric

equilibrium of hemoglobin toward a low oxygen affinity state. Moreover, the invention contemplates using the family of compounds for use in treating diseases involving oxygen deficiency, in wound healing, and in restoring oxygen affinity of stored blood.

Description of the Prior Art

Hemoglobin is a tetrameric protein which delivers oxygen via an allosteric mechanism. Oxygen binds to the four hemes of the hemoglobin molecule. Each heme contains porphyrin and iron in the ferrous state. The ferrous iron-oxygen bond is readily reversible. Binding of the first oxygen to a heme requires much greater energy than the second oxygen molecule, binding the third oxygen requires even less energy, and the fourth oxygen requires the lowest energy for binding. Hemoglobin has two α and two β subunits arranged with a two fold symmetry. The α and β dimers rotate during oxygen release to open a large central water cavity. The allosteric transition that involves the movement of the alpha-beta dimer takes place between the binding of the third and fourth oxygen. The $\alpha_1\beta_1$ interface binding is tighter than the $\alpha_1\alpha_2$ or $\alpha_1\beta_2$ interfaces.

In blood, hemoglobin is in equilibrium between two allosteric structures. In the "T" (for tense) state, hemoglobin is deoxygenated. In the "R" (for relaxed) state, hemoglobin is oxygenated. An oxygen equilibrium curve can be scanned, using well known equipment such as the AMINCO™ HEM-O-SCAN, to observe the affinity and degree of cooperativity (allosteric

action) of hemoglobin. In the scan, the Y-axis plots the percent of hemoglobin oxygenation and the X-axis plots the partial pressure of oxygen in millimeters of mercury (mm Hg). If a horizontal line is drawn from the 50% oxygen saturation point to the scanned curve and a vertical line is drawn from the intersection point of the horizontal line with the curve to the partial pressure X-axis, a value commonly known as the P_{50} is determined (i.e., this is the pressure in mm Hg when the scanned hemoglobin sample is 50% saturated with oxygen). Under physiological conditions (i.e., 37°C, pH=7.4, and partial carbon dioxide pressure of 40 mm Hg), the P_{50} value for normal adult hemoglobin (HbA) is around 26.5 mm Hg. If a lower than normal P_{50} value is obtained for the hemoglobin under test, the scanned curve is considered to be "left-shifted" and the presence of high affinity hemoglobin is indicated. Conversely, if a higher than normal P_{50} value is obtained for the hemoglobin under test, the scanned curve is considered to be "right-shifted" and the presence of low affinity hemoglobin is indicated.

It has been proposed that influencing the allosteric equilibrium of hemoglobin is a viable avenue of attack for treating diseases. The conversion of hemoglobin to a high affinity state is generally regarded to be beneficial in resolving problems with deoxy Hemoglobin-S (sickle cell anemia). The conversion of hemoglobin to a low affinity state is believed to have general utility in a variety of disease states where tissues suffer from low oxygen tension, such as ischemia and radio

sensitization of tumors. Several synthetic compounds have been identified which have utility in the allosteric regulation of hemoglobin and other proteins. For example, several new compounds and methods for treating sickle cell anemia which involve the allosteric regulation of hemoglobin are reported in U.S. Patent 4,699,926 to Abraham et al., U.S. Patent 4,731,381 to Abraham et al., U.S. Patent 4,731,473 to Abraham et al., U.S. Patent 4,751,244 to Abraham et al., and U.S. Patent 4,887,995 to Abraham et al. Furthermore, in both Perutz, "Mechanisms of Cooperativity and Allosteric Regulation in Proteins", *Quarterly Reviews of Biophysics* 22, 2 (1989), pp. 163-164, and Lalezari et al., "LR16, a compound with potent effects on the oxygen affinity of hemoglobin, on blood cholesterol, and on low density lipoprotein", *Proc. Natl. Acad. Sci., USA* 85 (1988), pp. 6117-6121, compounds which are effective allosteric hemoglobin modifiers are discussed. In addition, Perutz et al. has shown that a known antihyperlipoproteinemia drug, bezafibrate, is capable of lowering the affinity of hemoglobin for oxygen (see, "Bezafibrate lowers oxygen affinity of hemoglobin", *Lancet* 1983, 881. German Patent Applications 2,149,070 and 2,432,560, both to Witte et al., disclose compounds which are structurally similar to some of the compounds in the family of compounds defined by this invention. However, the Witte et al. patent applications contemplate use of the compounds for the reduction of serum lipid levels. The Witte et al. patent applications do not provide any indication of the potential use of the compounds for

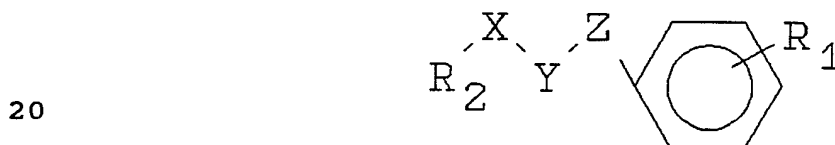
allosteric hemoglobin modification.

SUMMARY OF THE INVENTION

5 It is therefore an object of the present invention to provide a method of using a family of compounds to allosterically modify hemoglobin such that the hemoglobin is present in blood in a lower oxygen affinity state.

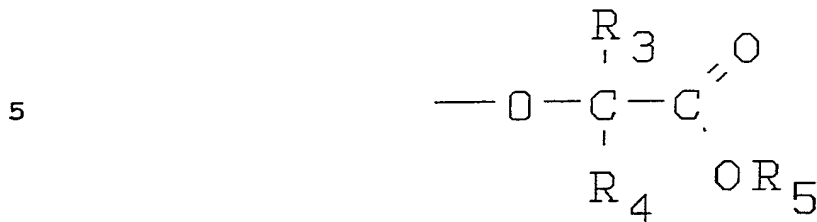
10 It is another object of the present invention to provide a method of prolonging the storage life of blood by adding compounds within a particular family of compounds to the blood.

15 According to the invention, an allosteric hemoglobin modifying family of compounds is defined by the formula:



where R_2 is a substituted or unsubstituted aromatic such as a phenyl, naphthyl, or indanyl, or a
25 heterocyclic aromatic, or a substituted or unsubstituted alkyl ring compound such as a cyclohexyl or adamantyl, or a substituted or unsubstituted phthalimide compound where X is a
30 carboxyl, Y is a nitrogen and R_2 completes the phthalimide compound by being bonded to both X and Y , and where X , Y , and Z are CH_2 , NH , CO , O or N with the caveat that the X , Y , and Z moieties are each different from one another, and where R_1 has

the formula:



10 where R1 can be connected to any position on the phenyl ring and R₃ and R₄ are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R₃ and R₄, and R₅ is a

15 hydrogen, loweralkyl such as methyl, ethyl or propyl, or a salt cation such as sodium, potassium, or ammonium. Many compounds within this family have been synthesized and their effect on the P₅₀ value of hemoglobin has been determined. Each of the

20 compounds tested was found to increase the P₅₀ value of hemoglobin; hence, the compounds are capable of driving the allosteric equilibrium of hemoglobin towards a condition favoring the low oxygen affinity state. In addition, the compounds were found to

25 stabilize the degree of oxygen dissociation of hemoglobin in stored blood over extended periods of time. Furthermore, the compounds were found to be well tolerated by mice when administered as an intraperitoneal dose. Because the compounds within

30 the family defined by this invention are capable of shifting the hemoglobin allosteric equilibrium toward the low affinity "T" state, they have the ability to cause hemoglobin to deliver more oxygen

to tissues. Thus, the compounds of the invention should be valuable as antiischemic agents, as sensitizers for x-ray irradiation in cancer therapy, as wound healing agents, in treating disorders
5 related to low oxygen delivery in the brain such as Alzheimer's, depression, and schizophrenia, in preparing blood substitutes, and in blood storage.

BRIEF DESCRIPTION OF THE DRAWINGS

10

The foregoing and other objects, aspects and advantages will be better understood from the following detailed description of a preferred embodiment of the invention with reference to the
15 drawings, in which:

Figure 1a is a chemical structure defining a particularly preferred group within the family of compounds used in the present invention;

20 Figures 1b and 1c are chemical structures defining two subsets of the family defined in Figure 1a;

Figures 2a-b depict chemical structures of precursor compounds arranged in reaction schemes for preparing compounds that are useful as intermediates
25 for synthesizing compounds within a first group of the family of compounds;

Figure 2c depicts chemical structures, including the intermediates produced as shown in Figures 2a-b, arranged in a reaction scheme to
30 prepare the first group of preferred compounds;

Figure 3 depicts chemical structures arranged in a reaction scheme to produce a second group of the family of preferred compounds;

Figure 4 depicts chemical structures arranged in a reaction scheme to produce a third group of the family of preferred compounds;

5 Figure 5a-b depict chemical structures of precursor compounds arranged in reaction schemes for preparing compounds that are useful as intermediates for synthesizing compounds within a fourth group of the family of preferred compounds;

10 Figure 5c depicts chemical structures, including the intermediates produced in Figures 5a-b, arranged in a reaction scheme to produce the fourth group of compounds;

15 Figure 6a depicts chemical structures arranged in a reaction scheme, which is an alternative to that shown in Figure 4, for producing compounds within a third group of the family of preferred compounds;

20 Figure 6b depicts chemical structures arranged in a reaction scheme similar to that shown in Figure 6a, except that the precursor compounds utilized are chosen such that the compound produced has a meta-substitution rather than para-substitution on one phenyl ring and so that ethyl rather than methyl groups are present on the substituent moiety of the meta-substituted phenyl ring;

25

Figures 7a and 7b depict chemical structures arranged in a reaction scheme for producing compounds within a fifth group of the family of preferred compounds;

30 Figure 8 is a table presenting the measured P_{50} values for hemoglobin in solution where the addition of each of the compounds within the preferred family was shown to allosterically modify hemoglobin

towards the low oxygen affinity state;

Figure 9 is a table similar to that shown in Figure 8a except that the measured P_{50} values are for intact human red blood cells (as opposed to in hemoglobin solution) exposed to some of the compounds with the family defined by this invention;

Figure 10 is a graph illustrating the oxygen dissociation curves produced when a 5.4 millimolar solution of normal hemoglobin in the presence and absence of selected compounds is tested at pH 7.4 using HEPES as the buffer in a Hem-O-Scan oxygen dissociation analyzer;

Figure 11 is a graph similar to Figure 10 which illustrates oxygen dissociation curves for whole human blood in the presence and absence of selected compounds;

Figure 12 is a graph similar to Figure 10 where the oxygen dissociation curves produced are for a 5.4 millimolar solution of normal hemoglobin in the presence and absence of particular compounds, including 2,3-diphosphoglycerate which is the natural allosteric hemoglobin effector, are tested at pH 7.4 using HEPES as the buffer in a Hem-O-Scan oxygen dissociation analyzer;

Figure 13 is a table showing the effect of 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{20}H_{23}NO_4$) on human red blood cells stored in adsol formulation;

Figure 14 is a bar graph showing the percentage oxygen delivered by packed cells, fresh stored and in the presence of 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{20}H_{23}NO_4$), respectively;

Figure 15 is a table showing the change in the P_{50} values of outdated packed red blood cells on treatment with 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid (called RSR-13); and

Figure 16 depicts chemical structures in a reaction scheme used to produce a phthalimide form of the compounds within the present invention where the compounds were shown by a measured P_{50} value to allosterically modify hemoglobin towards the low oxygen affinity state.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS OF THE INVENTION

Referring now to the drawings and, more particularly, to Figures 1a-c which illustrate the general structural formula of particularly preferred compounds contemplated for use in the present invention and first and second subsets of the general structural formula, respectively. With reference to the general structural formula of Figure 1a, the X and Z moieties may be CH_2 , CO, NH or O, and the Y moiety may be CO or NH, with the caveat that the X, Y, and Z moieties are each different from one another. In addition, R_{2-6} are either hydrogen, halogen, a substituted or unsubstituted C_{1-3} alkyl group (up to three carbons in length), or a C_{1-3} ester or ether and these moieties may be the same or different, or alkyl moieties of aliphatic or aromatic rings incorporating two adjacent R_{2-6} sites. The R_{7-8} positions are hydrogen, halogen, methyl, or ethyl

groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic (e.g., cyclobutyl) ring connecting R₇ and R₈. The R₉ position is a hydrogen, halogen, C₁₋₃ loweralkyl such as methyl, ethyl or propyl, or a salt cation such as sodium, potassium, or ammonium.

In the first subset of compounds defined in Figure 1b, X and Z may each be CH₂, NH, or O, with the caveat that when X is CH₂, Z is either NH or O, when X is NH, Z is either CH₂ or O, and when X is O, Z is NH or CH₂. The first subset of compounds may conveniently be classified into four groupings as follows:

Group I: 2-[4-((aryl)acetamido)phenoxy]-2-methyl propionic acid

compounds having the general structural formula illustrated in Figure 2C;

Group II: 2-[4-(((aryl)oxy)carbonyl)amino)phenoxy]-2-methyl propionic acid compounds having the general structural formula illustrated in Figure 3;

Group III: 2-[4-(((aryl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid compounds having the general structural formula illustrated in Figures 4 and 6a; and

Group IV: 2-[4-(((aryl)amino)carbonyl)oxy)phenoxy]-2-methyl propionic acid compounds having the general structural formula

illustrated in Figure 5C.

In the second subset of compounds defined in Figure 1c, X and Z may each be CO or CH₂, with the caveat that when X is CO, Z is CH₂, and when X is CH₂, Z is CO. The second subset of compounds may be conveniently divided into two groupings as follows:

Group V: 2-[4-(((aryloyl)amino)methyl)phenoxy]-2-methylpropionic acid compounds having the general structural formula illustrated in Figure 7b.

Group VI: 2-[4-(((aryl)methyl)amino)carbonyl)phenoxy]-2-methylpropionic acid compounds which are the subject matter of the co-pending U.S. Patent application 07/623,346 to Abraham et al. filed December 7, 1990.

The R₂₋₉ substituents in Figures 1b-c are the same as those defined with reference to Figure 1a. The synthesis of specific chemical compounds within the first five groups of compounds is provided in the following examples with reference to Figures 2-7. The synthesis of specific chemical compounds in the sixth group is explained in detail in co-pending U.S. Patent application 07/623,346 to Abraham which has been incorporated by reference. All compounds which were prepared were checked by thin layer chromatography (TLC) for purity and the structure elucidation was based on NMR and IR spectroscopy and elemental analysis.

EXAMPLE 1

Figure 2A illustrates a reaction scheme for preparing 2-(4-aminophenoxy)-2-methyl propionic acid, a compound that is useful as a precursor in the preparation of Group I compounds. In accordance with the scheme of Figure 2A, 8 grams (g) (0.2 mol) of pulverized sodium hydroxide is added to a suspension of 5.28 g (0.035 mol) of p-acetaminophenol in 23 g (0.4 mol) of acetone. The reaction mixture is stirred at room temperature for 1/2 hour. Subsequently, 3.58 g (0.03 mol) of chloroform is added dropwise over the course of 30 minutes. The reaction mixture is stirred overnight at room temperature and acetone is removed under vacuum. The residue is dissolved in water (10 ml), followed by acidification with 37% hydrochloric acid (HCl) to produce a pale yellow precipitate of 2-(4-acetaminophenoxy)-2-methyl propionic acid (5 g, 60% yield), crystallized from methanol, mp 69-71°C.

^1H NMR : (CD₃OD) δ 7.1(m,4H) ArH, 2.05 (s,3H), CH₃, 1.45, (s,6H) 2CH₃

1.18 g (0.005 mol) of the 2-(4-acetaminophenoxy)-2-methyl propionic acid is boiled in 10% KOH (60 ml) for 1/2 hour. The solution is then cooled and acidified with acetic acid to yield 0.6 g (62% yield) of 2-(4-aminophenoxy)-2-methyl propionic acid as a yellowish white powder, mp 214-16°C.

^1H NMR: (DMSOd₆+TMS) δ 6.6 (m,4H)ArH, 1.35 (s,6H, 2CH₃)

EXAMPLE 2

Figure 2B illustrates another reaction scheme for preparing 2-(4-aminophenoxy)-2-methyl propionic acid. In accordance with the scheme of Figure 2B, 8 grams of potassium hydroxide is dissolved in 32 ml of water and the resultant KOH solution is admixed with 280 ml of 3% hydrogen peroxide. 11.3 g (0.058 mol) of 2-(4-cyanophenoxy)-2-methyl propionic acid is slowly added to the KOH/H₂O₂ solution and the reaction mixture is stirred for about one hour until the exotherm and evolution of gas has ceased. The mixture is then cooled and acidified with concentrated hydrochloric acid. The 2-[4-(carboxamido)phenoxy]-2-methyl propionic acid product is obtained as a white solid (9.8 g, 79% yield). The product is crystallized from ethanol to produce pure white crystals, mp 202-4°C.

5.57 g (0.025 mol) of the 2-[4-(carboxamido)phenoxy]-2-methyl propionic acid is added gradually with stirring to 100 ml of an ice cold aqueous solution containing 4.4 g (0.025 mol) of bromine and 11 g (0.25 mol) of sodium hydroxide. The solution thus obtained is warmed at 75°C for 1/2 hour. After cooling, the solution is acidified with acetic acid giving the desired 2-(4-aminophenoxy)-2-methyl propionic acid product as 4.0 g (81% yield) of a white precipitate, mp 214-16°C. The compound is identical with the product prepared in Example 1.

EXAMPLE 3

Figure 2C illustrates a general reaction scheme for preparing the Group I 2-[4-(arylacetamido)phenoxy]-2-methyl propionic acids. In accordance with the illustrated scheme, 1 g (0.005 mol) of 2-(4-aminophenoxy)-2-methyl propionic acid is dissolved with stirring in 10 ml of water containing 0.41 g (0.1 mol) of NaOH. To this solution, 0.79 g (0.005 mol) of phenyl acetyl chloride in 5 ml of tetrahydrofuran (THF) is gradually added over a period of about 15 minutes. After the addition is complete the pH of the reaction mixture should be alkaline (if not a few drops of 2N NaOH is added to assure alkalinity). The reaction mixture is continuously stirred for 1 hour. Thereafter, the THF is evaporated in vacuo, and the solution is then diluted with 5 ml water and acidified with concentrated hydrochloric acid. The product is extracted with ethyl ether (2 X 20 ml), washed with water (3 X 20 ml), and then dried over anhydrous MgSO_4 . Upon addition of petroleum ether to the ether solution, 0.9 g (56% yield) of the 2-[4-(phenylacetamido)phenoxy]-2-methyl propionic acid product precipitates as a pale brown solid, mp 173-175°C.

^1H NMR: (DMSO-d₆) 10 (s, 1H, COOH), 7.5-6.7 (m, 9H, ArH), 3.55 (s, 2H, CH₂), 1.4 (s, 6H, 2CH₃)

30

Anal: $\text{C}_{18}\text{H}_{19}\text{NO}_4$

Calculated C 69.00 H 6.07 N 4.47

Found C 68.86 H 6.14 N 4.42

EXAMPLE 4

The procedure of Example 3 is followed as above, except that 0.005 mol of 4-chlorophenyl acetyl chloride is substituted for the phenyl acetyl chloride. In this case the product (57% yield) is 2-[4-(p-chlorophenylacetamido)phenoxy]-2-methyl propionic acid, mp 168-71°C.

¹H NMR: (DMSOd6) δ 10 (s, 1H, COOH), 7.6-6.7 (m, 8H, ArH), 3.6 (s, 2H, CH₂), 1.4 (s, 6H, 2CH₃)

Anal: C₁₈H₁₈NO₄Cl

Calculated C 62.15 H 5.17 N 4.02 Cl 10.12

Found C 62.16 H 5.25 N 3.98 Cl 10.25

The 4-chlorophenyl acetyl chloride for the foregoing synthesis is prepared by heating to reflux a suspension of 1 g (0.006 mol) of 4-chlorophenyl acetic acid in 1.07 g (0.009 mol) of thionyl chloride with stirring for 1 hour. After cooling, excess thionyl chloride is evaporated under vacuum to present the 4-chlorophenyl acetyl chloride product as a yellow oil (1 g, 83% yield).

EXAMPLE 5

Figure 3 illustrates a general reaction scheme for preparing the Group II 2-[4-(((aryloxy)carbonyl)amino)phenoxy]-2-methyl propionic acids. In accordance with the illustrated scheme, a solution consisting of 0.15 g (0.001 mol) of phenyl chloroformate in 3 ml THF is gradually added to an ice cold solution containing 0.3 g (0.001 mol) of 2-(4-amino phenoxy)-2-methyl

propionic acid and 0.17 g (0.002 mol) of sodium bicarbonate in 10 ml of water (10 ml). The reaction mixture is stirred for 1/2 hour at 0°C, followed by stirring for 1 hour at room temperature. The THF is removed in vacuo and 10 ml of water is added. Then, the reaction mixture is acidified with concentrated hydrochloric acid and extracted with ethyl ether (2 X 20 ml). The ether solution is washed with water (3 X 20 ml) and dried over anhydrous MgSO₄. The desired product, 2-[4-(((phenyl)oxy)carbonyl)amino)phenoxy]-2-methyl propionic acid, is precipitated from the ether solution by addition of petroleum ether as a white solid, 0.15 g (31% yield), mp 183-5°C.

¹H NMR: (DMSOd₆) δ 10 (s, 1H, COOH), 7.55-6.75 (m, 9H, ArH), 1.4 (s, 6H, 2CH₃)

Anal: C₁₇H₁₇O₅N

Calculated C 64.76 H 5.39 N 4.44

Found C 64.65 H 5.45 N 4.43

EXAMPLE 6

The procedure for Example 5 is followed as above except that 0.001 mol of 4-chlorophenyl chloroformate is substituted for the phenyl chloroformate. In this case the 2-[4-(((p-chlorophenyl)oxy)carbonyl)amino)phenoxy]-2-methyl propionic acid product is obtained as a white precipitate, 0.15 g (28% yield), mp 179-82°C.

¹H NMR: (DMSOd₆+TMS) δ 7.6-6.8 (m, 8H, ArH), 1.4 (s, 6H, 2CH₃)

Anal: C₁₇H₁₆O₅NCl

Calculated C 58.36 H 4.57 Cl 10.15

Found C 58.16 H 4.68 Cl 10.35

EXAMPLE 7

5 Figure 4 illustrates a general reaction scheme
for preparing the Group III compounds of the
invention. In accordance with the illustrated
scheme, 5.2 g (34 mmol) of 4-hydroxyphenylacetic
10 acid (HPAA) is heated to reflux with an excess of
thionyl chloride (SOCl_2) for 1/2 hour. The reaction
mixture is then cooled and excess SOCl_2 is removed
under vacuum. The residue is reacted for 2 hours
with 6.3 g (68 mmol) of aniline in 50 ml of
15 refluxing xylene. The reaction mixture is then
cooled, washed with dilute HCl, water and brine and
extracted with aqueous 2N NaOH. The combined alkali
layer is washed with ether, cooled and acidified to
provide 7 g of solid N-phenyl-4-hydroxybenzyl amide
20 ($\text{C}_{14}\text{H}_{12}\text{NO}_2$) as an intermediate product (90% yield), mp
138°C. The intermediate product is recrystallized
from a 1:2 mixture of acetone and petroleum ether
and a 1.13 g (5 mmol) portion is O-alkylated for 12
hours using the procedure of Example 1 with 20 ml
acetone, 2.75 g NaOH and 1.25 ml CHCl_3 . The final
25 product is 2-[4-(((phenyl)amino)carbonyl)
methyl]phenoxy]-2-methyl propionic acid ($\text{C}_{18}\text{H}_{19}\text{NO}_4$),
1.2 g (76% yield), mp 198°C.

EXAMPLE 8

30

The procedure of Example 7 is repeated using 8.6
g (68 mmol) of 4-chloroaniline rather than the
aniline. In this case, the intermediate product is

N-(4-chlorophenyl)-4-hydroxy benzylamide
($C_{14}H_{12}ClNO_2$), 7.5 g (84% yield), mp 163°C. 1.3 g
of the intermediate product is O-alkylated to produce
2-[4-(((4-chlorophenyl)amino)carbonyl)methyl)
5 phenoxy]-2-methyl propionic acid ($C_{18}H_{18}ClNO_4$),
0.86 g (50% yield), mp 196°C.

EXAMPLE 9

10 The procedure of Example 7 is repeated using
2.6 g (17 mmol) of the HPAA and using 5.67 g (35
mmol) of 3,4-dichloroaniline rather than aniline.
In this case, the intermediate product is N-(3,4-
dichlorophenyl)-4-hydroxy benzylamide ($C_{14}H_{11}Cl_2NO_2$).
15 1.48 g (5 mmol) of the intermediate is O-alkylated
to produce 2-[4-(((3,4-dichlorophenyl)amino)
carbonyl)methyl)phenoxy]-2-methyl propionic acid
($C_{18}H_{17}Cl_2NO_4$), 0.76 g (40% yield), mp 174°C.

20

EXAMPLE 10

The procedure of Example 7 is repeated using
2.6 (17 mmol) of the HPAA and using 5.7 g (35 mmol)
of 3,5-dichloroaniline rather than aniline. In this
25 case, the intermediate product is N-(3,5-
dichlorophenyl)-4-hydroxy benzylamide ($C_{14}H_{11}Cl_2NO_2$).
1.48 g (5 mmol) of the intermediate is O-alkylated
to produce 2-[4-(((3,5-dichlorophenyl)amino)
carbonyl)methyl)phenoxy]-2-methyl propionic acid
30 ($C_{18}H_{17}Cl_2NO_4$), 0.8 g (42% yield), mp 138°C.

EXAMPLE 11

The procedure of Example 7 is repeated using 0.95 g (6 mmol) of the HPAA, 2.6 g (12 mmol) of 3,4,5-trichloroaniline rather than aniline, and 25 ml of refluxing xylene. In this case, the intermediate product is N-(3,4,5-trichlorophenyl)-4-hydroxy benzylamide. 0.50 g (1.5 mmol) of the intermediate product is O-alkylated using 10 ml acetone, 0.82 g NaOH and 0.37 ml CHCl_3 to produce 2-[4-(((3,4,5-trichlorophenyl)amino)carbonyl)methyl]phenoxy]-2-methyl propionic acid ($\text{C}_{18}\text{H}_{16}\text{Cl}_3\text{NO}_4$), 0.27 g (43% yield), mp 160°C.

EXAMPLE 12

The procedure of Example 7 is repeated using 5.04 g (32 mmol) of the HPAA, 6 ml (64 mmol) of 4-fluoroaniline rather than aniline, and 25 ml of refluxing xylene. In this case, the intermediate product is N-(4-fluorophenyl)-4-hydroxybenzylamide. 1.22 g (5 mmol) of the intermediate product is O-alkylated to produce 2-[4-(((4-fluorophenyl)amino)carbonyl)methyl]phenoxy]-2-methyl propionic acid ($\text{C}_{18}\text{H}_{18}\text{FNO}_4$), 0.74 g (45% yield), mp 198°C.

EXAMPLE 13

The procedure of Example 7 is repeated using 5.04 g (32 mmol) of the HPAA, 8.05 ml (64 mmol) of 4-trifluoromethylaniline rather than aniline, and 25 ml of refluxing xylene. In this case, the

intermediate product is N-(4-trifluoromethylphenyl)-4-hydroxy benzylamide. 1.5 g (5 mmol) of the intermediate is used to produce 2-[4-(((4-trifluoromethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{19}H_{18}F_3NO_4$), 0.85 g (44% yield), mp 197°C.

EXAMPLE 14

The procedure of Example 7 is repeated using 5.04 (32 mmol) of the HPAA, 8 g (65 mmol) of 4-methyl aniline rather than aniline, and using 25 ml of refluxing xylene. In this case the intermediate product is N-(4-methylphenyl)-4-hydroxy benzylamide. 1.2 g (5 mmol) of the intermediate is used to produce 2-[4-(((4-methylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{19}H_{21}NO_4$), 0.98 g (65% yield), mp 164°C.

EXAMPLE 15

The procedure of Example 7 is repeated using 3.26 (21 mmol) of the HPAA, 5.3 ml (42 mmol) of 3,5-dimethyl aniline rather than aniline, and 25 ml of refluxing xylene. In this case the intermediate product is N-(3,5-dimethylphenyl)-4-hydroxy benzylamide. 1.27 g (5 mmol) of the intermediate is used to produce 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{20}H_{23}NO_4$), 1.15 g (68% yield), mp 85°C. Alternatively, the procedure outlined in the German Patent Application 2,432,560, which is herein

incorporated by reference, can be followed to produce the compound of this Example 15.

EXAMPLE 16

5 The procedure of Example 7 is repeated using
5.04 (32 mmol) of the HPAA, 10 ml (64 mmol) of
4-isopropyl aniline rather than aniline, and using
25 ml of refluxing xylene. In this case the
10 intermediate product is N-(4-isopropylphenyl)-4-
hydroxybenzylamide. 1.34 g (5 mmol) of the semi-
solid, thick viscous liquid intermediate is used to
prepare 2-[4-(((4-isopropylphenyl)amino)carbonyl)
methyl)phenoxy]-2-methyl propionic acid ($C_{21}H_{25}NO_4$),
15 1.1 g (61% yield), mp 141°C.

EXAMPLE 17

20 With reference to Figures 5A, 5B and 5C, a
scheme is illustrated for preparing Group IV
compounds. In accordance with Figure 5A, aniline or
aniline derivatives may be reacted with phosgene to
obtain the carbamoyl chloride. In accordance with
Figure 5B, hydroquinone may be monoacetylated using
25 acetic anhydride. The product is then O-alkylated
using acetone, $CHCl_3$ and KOH and then hydrolyzed
using a base. The products of the reactions of
Figures 5A and 5B may then be reacted according to
the reaction scheme of Figure 5C to produce the
30 Group IV 2-[4-(((arylamino)carbonyl)oxy)phenoxy)]-2-
methyl propionic acids.

EXAMPLE 18

As an alternative to the reaction scheme described in Example 7 and shown in Figure 4, the Group III compounds may be prepared according to the scheme shown in Figure 6a. 5.2 g (32 mmol) of HPAA, 6.3 g (68 mmol) of aniline, and 25 ml of mesitylene are heated to reflux. 0.74 g (8mmol) of phosphorous pentachloride is added to the refluxing mixture and the reflux is continued for an additional two hours. The reaction mixture is subsequently cooled, washed with dilute HCl, water and brine, and extracted with aqueous 2N sodium hydroxide NaOH. The combined alkali layer is washed with ether, cooled and acidified to provide 7 g (90% yield) of solid N-phenyl-4-hydroxybenzyl amide ($C_{14}H_{12}NO_2$) as an intermediate product, mp 138°. The intermediate product is recrystallized from a 1:2 mixture of acetone:petroleum ether and a 1.13 g (5 mmol) portion is O-alkylated. 1.6 g (30 mmol) of pulverized sodium hydroxide is added to a solution of N-phenyl-4-hydroxybenzamide (1.13g, 5mmol) in 20 ml of acetone.. The reaction mixture is stirred overnight at room temperature and acetone is removed under vacuum. The residue is dissolved in 10 ml of water and acidified with 2N HCl to produce a pale yellow solid. The solid is separated, dissolved in methanol, charcoaled, and solvent evaporated to provide 2-[4-(((phenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{18}H_{19}NO_4$), 1.2 g (76% yield), mp 198°C. The last step in the procedure shown in Figure 6a is the conversion of the acid to the sodium salt via its reaction with

sodium bicarbonate. Similar reactions with other salt cations such as potassium and ammonium or reactions to form esters can also be performed.

5

EXAMPLE 19

Figure 6b presents a similar reaction scheme to Figure 6a, except that 3- rather than 4-hydroxyphenylacetic acid (HPAA) is used as the precursor material so that the final compound has a meta rather than a para substitution. In addition, rather than reacting with acetone (dimethyl ketone) a diethyl ketone is used to position ethyl, rather than methyl, moieties in the group substituted on one of the phenyl rings. By example, 1.5g (10mmol) 3-HPAA and 2.6g (20 mmol) 4-chloroaniline in 20 ml of mesitylene was heated to reflux. Then 0.33g (2.55 mmol) PCl_5 solution was then slowly added to the above refluxing solution and the refluxing was continued for two hours. The reaction mixture was then cooled and then worked up as described above to yield 2.2 g (90% yield) of 3-(((4-chloroanilino)carbonyl)methyl]phenol. As described above, chloroform (0.8 ml) was added to a stirred and ice-cooled mixture of 1.23 grams of 3-(((4-chloroanilino)carbonyl)methyl]phenol and 1.6 g NaOH in 15 ml of acetone. The reaction mixture was allowed to warm to room temperature and stirring continued for an additional 10 hours. The usual work-up yielded 2-[3-(((4-chloroanilino)carbonyl)methyl]phenoxy]-2-methylpropionic acid as a low temperature melting sticky solid (C,H,Cl,N analysis yielded $(\text{C}_{18}\text{H}_{18}\text{ClNO}_4)$; NMR δPPM : 1.42 (6H, s, CH_3),

3.61 (2H, s, benzylic CH₂), and 6.6-7.75 (8H, m, aromatic H)). However, rather than using acetone as the reaction solvent, diethylketone can be used in the same manner as described above to yield the butanoic acid (as opposed to propanoic acid) structure shown in Figure 6b.

EXAMPLE 20

With reference to Figures 7a, a general reaction scheme for preparing 2-[4-(aminomethyl)phenoxy]-2-methyl propionic acid, a compound that is useful as a precursor to the preparation of the Group V compounds, is presented. In accordance with the illustrated scheme, 2-[4-cyanophenoxy]-2-methyl propionic acid (2g, 9mmol), prepared as described in Example 2, and 75 ml of ethanol were placed in a 250 ml Parr hydrogenation bottle. The solution was acidified with concentrated hydrochloric acid (3 ml), then 10% palladium on activated charcoal (0.2 g, 10% wt) was added to the mixture. The reaction mixture was placed on a Parr hydrogenator apparatus at 45 psi of hydrogen pressure and shaken for a period of two hours. The mixture was filtered to remove the catalyst, and the filtrate concentrated under vacuum. Addition of ether precipitated hydrochloride salt of the desired product as white, shiny crystals (2.1 g, 87%).

EXAMPLE 21

Figure 7B illustrates a general reaction scheme for preparing the Group V compounds used in the present invention. In accordance with the illustration, a solution of benzoyl chloride (0.14g, 1mmol) in THF (3ml) was added over a 15 minute period to a stirred solution of 2-[4-(aminomethyl)phenoxy]-2-methylpropionic acid (0.24g, 1 mmol) and NaOH (0.08g, 2 mmol) in 10 ml of water. After the addition of the benzoyl chloride was completed, the reaction mixture was stirred for 1 hour at room temperature. THF was evaporated in vacuo. Acidification of the residue provided the desired compound as an oil which was extracted with ether. The organic layer was washed with water, brine, and aired over anhydrous MgSO_4 . Subsequent addition of petroleum ether precipitated 2-[4-(benzoylamino)methyl]phenoxy]-2-methyl propionic acid ($\text{C}_{18}\text{H}_{19}\text{NO}_4$) as a white solid (0.15g, 48%) mp 176-179°C.

NMR: (DMSO-d_6) δ 1.45 (6H, s, 2CH_3), 4.4 (2H, d, CH_2), 6.8-7.2 (4H, dd, $J=9$ Hz, aromatic H), 7.4-8 (5 H, m, aromatic H), 9, 1 H, br t, NH).

EXAMPLE 22

The procedure of Example 21 is repeated using 2-chlorobenzoyl chloride (1 mmol) rather than benzoyl chloride. In this case, the product (58% yield) is 2[4-(((2-chlorobenzoyl)amino)methyl)phenoxy]-2-methyl propionic acid ($\text{C}_{18}\text{H}_{18}\text{ClNO}_4$) mp

135-137°C.

EXAMPLE 23

5 The procedure of Example 21 is repeated, except
that 1 mmol of 3-chlorobenzoyl chloride is
substituted for benzoyl chloride. In this case, the
product (53% yield) is 2-[4-(((3-chlorobenzoyl)
10 amino)methyl)phenoxy]-2-methyl propionic acid
(C₁₈H₁₈ClNO₄) mp 145-146°C.

EXAMPLE 24

15 The procedure of Example 21 is repeated, except
that 1 mmol of 4-chlorobenzoyl chloride is
substituted for benzoyl chloride. In this case, the
product (63% yield) is 2-[4-(((4-chlorobenzoyl)
amino)methyl)phenoxy]-2-methyl propionic acid
20 (C₁₈H₁₈ClNO₄) mp 186-189°C.

EXAMPLE 25

25 The procedure of Example 21 is repeated, except
that 1 mmol of 3,4-dichlorobenzoyl chloride is
substituted for benzoyl chloride. In this case, the
product (57% yield) is 2-[4-(((3,4-dichlorobenzoyl)
amino)methyl)phenoxy]-2-methyl propionic acid
30 (C₁₈H₁₇Cl₂NO₄) mp 186-189°C.

EXAMPLE 26

30 The procedure of Example 20 is repeated, except
that 1 mmol of 3,5-dichlorobenzoyl chloride is

substituted for benzoyl chloride. In this case, the product (43% yield) is 2-[4-(((3,5-dichlorobenzoyl) amino)methyl)phenoxy]-2-methyl propionic acid ($C_{18}H_{17}Cl_2NO_4$) mp 110-113°C.

5

EXAMPLE 27

The procedure of Example 20 is repeated, except that 1 mmol of 3,4,5-trichlorobenzoyl chloride is substituted for benzoyl chloride. In this case, the product is 2-[4-(((3,4,5-trichlorobenzoyl) amino)methyl)phenoxy]-2-methyl propionic acid ($C_{18}H_{16}Cl_3NO_4$) mp 151-152°C.

10

Examples 1 through 27 outline the synthesis procedures for producing several compounds within the family of compounds defined by the general structural formula of Figure 1a. Specifically, Examples 1-19 disclose synthesis procedures for Groups 1-4 compounds within the subset defined by the structural formula of Figure 1b and Examples 20-27 disclose synthesis procedures for Group 5 compounds within the subset defined by the structural formula of Figure 1c. The co-pending U.S. Patent application 07/623,346 to Abraham et al. filed December 7, 1990, describes the synthesis procedures for Group 6 compounds within the subset defined by the structural formula of Figure 1c. It should be understood that other compounds within the family of compounds used in the present invention can easily be synthesized by changing the starting materials. All compounds within the family would have a similar mode of binding and would, therefore, all should have the effect of shifting the

15

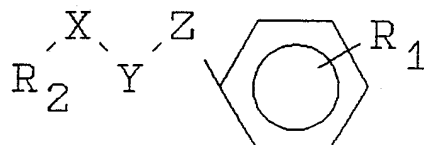
20

25

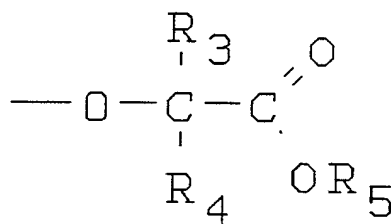
30

allosteric equilibrium of hemoglobin towards favoring the low affinity "T" state.

The broad family of compounds contemplated for use in this invention includes compounds defined by the formula:



where R_2 is a substituted or unsubstituted aromatic such as a phenyl, naphthyl, or indanyl, or hetrocyclic aromatic, or a substituted or unsubstituted alkyl ring compound, such as a cyclohexyl or adamantyl, or a substituted or unsubstituted phthalimide compound where X is a carboxyl, Y is a nitrogen and R_2 completes the phthalimide compound by being bonded to both X and Y, and where X, Y, and Z are CH_2 , NH, CO, O or N with the caveat that the X, Y, and Z moieties are each different from one another, and where R_1 has the formula:



where R_1 can be connected to any position on the phenyl ring and R_3 and R_4 are hydrogen, halogen,

methy1, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R₃ and R₄, and R₅ is a hydrogen, loweralkyl such as methyl, ethyl or propyl, or a salt cation such as sodium, potassium, or ammonium. To this end, compounds have having a naphthyl, adamantyl, or indanyl group at R₁ instead of the substituted phenyl like that shown in Figure 1a have been prepared using substantially the same synthetic routes as described above. In addition, compounds having a phthalimide-like structure have also been synthesized as shown in Figures 16a-b and described below in EXAMPLE 28.

15

EXAMPLE 28

Figures 16a and 16b show alternative synthesis routes for preparing 2-[4-((phthalamido)N-methyl)phenoxy]-2-methyl proprionic acid. Phthalic anhydride (0.75g; 5mmol) and 2[4-((amino)methyl)phenoxy]-2-methyl proprionic acid (see Figure 2B) were refluxed in 25 ml of toluene in the presence of 1ml triethylamine. Water was removed azeotropically. After four hours of refluxing, the reaction mixture was cooled, toluene was separated, and the above-described work up was provide to yield a crystalline white residue (90% yield; mp 149°C; NMR δ ppm: 1.46(6H, s, CH₂), 4.65(2H, s, CH₂), 6.75 and 7.2 (4H, d, J = 6Hz, aromatic H of a para-substituted ring), and 7.85 (4H, s, aromatic H of a phthalimide unit).

30

To test the compounds of the invention for physiological activity, human blood was obtained

from the Central Blood Bank, Richmond, Virginia. The extraction, chromatography, and characterization of isolated hemoglobin methods used by the inventors were identical to those described by Dozy and Huisman in J. of Chromatography, Vol 32, (1968) pp. 723 and in The Chromatography of Hemoglobin, H.J. Schroeder and D.H.J. Huisman, Ed. Marcel Dekker Inc. N.Y. (1980) which are herein incorporated by reference. The purity of normal hemoglobin (HbA) was determined by gel electrophoresis, using a Gelman semimicroelectrophoresis chamber. The concentration of hemoglobin was determined according to the cyanmethemoglobin method described in Zijlstra, Clin. Chem. Acta., Vol 5, pp. 719-726 (1960), and Zijlstra and Van Kamper, J. Clin. Chem. Clin. Biochem., Vol. 19, p. 521 (1981) which are herein incorporate by reference. All purified hemoglobin solutions were stored in liquid nitrogen. The reagents and buffers were purchased from the following sources: Fischer Scientific, Sigma Chemical Company, and Pharmacia and Research Chemicals, Inc.

Oxygen equilibrium curves were determined on an AMINCO™ HEM-O-SCAN oxygen dissociation analyzer available from Travenol Laboratories. HbA was prepared as follows: 20 ml of whole blood from a nonsmoking donor (blood bank, Richmond, Virginia) was drawn into a heparinized vacutainer. The blood was immediately packed in ice (to prevent Methb formation) and then centrifuged (10 minutes at 2500 rpm) to separate the plasma and buffy coat from the packed erythrocytes. After centrifugation was completed, the plasma and buffy coat were removed by

aspiration and the cells washed three times with 0.9% NaCl containing 40 mg of ethylenediamine-tetraacetic acid (EDTA) per liter and then once with 1.0% NaCl containing 40 mg of EDTA/L. The cells
5 were lysed by the addition of one to two volumes of deionized water containing 40 mg of EDTA/L. The mixture was allowed to stand for 30 minutes with occasional mixing before being centrifuged for two hours at 10,000 rpms at 4°C for two hours to remove
10 the remaining cell stroma. The supernatant was further purified by either gel filtration with Sephadex G-25 or dialysis against pH 8.6 tris buffer (50 mM, containing 40 mg. of EDTA/L). The sodium chloride free hemoglobin solution was
15 chromatographed on DEAE-Sephacel ion-exchange resin (Sigma) preequilibrated with Tris buffer (pH 8.6, 50 mM, containing 40 mg of EDTA/L), the HbA fraction was then eluted with pH 8.4 Tris buffer. The pure HbA fraction (identified by electrophoresis) was
20 concentrated using a Schleicher and Schuell collodion bag apparatus (Schleicher and Schuell, Inc.) with HEPES buffer (150 mM, pH 7.4) as the exchange buffer. The hemoglobin concentration was then determined using the above-noted
25 cyanomethemoglobin method. The hemoglobin concentration at this point was usually found to be around 35g% or approximately 5.5mM. Less than 5% methemoglobin was noted even after several days at 4°C.

30 All compounds were mixed with one equivalent of sodium bicarbonate (NaHCO_3) (this process converts the carboxylic acid moiety to a sodium salt; see Fig. 6a), then dissolved in the HEPES buffer to give

20 mM solutions. Just prior to running the oxygen equilibrium curve, the hemoglobin and the drug were mixed in a 1:1 ratio (50 μ l of hemoglobin plus 50 μ l of drug) to give 2.75 mM hemoglobin with a drug concentration of 10 mM. The control was prepared by the addition of 50 μ l of hemoglobin to 50 μ l of the HEPES buffer.

Figure 8 presents the measured P_{50} value, the P_{50} control value, and the ratio of the measured P_{50} value to the control (P_{50}/P_{50C}) for normal hemoglobin treated with several synthesized compounds. It is noted that the P_{50} control value is less than for normal hemoglobin under physiological conditions (e.g., 26.5) because here the P_{50} was made on hemoglobin in solution (outside the red blood cells). Each hemoglobin sample treated with one of the compounds falling within the family defined by this invention had a P_{50} drug value which was greater than the P_{50} control. This response indicates that the allosteric equilibrium for hemoglobin has been shifted towards favoring the low oxygen affinity "T" state of hemoglobin due to the presence of the compounds. At the bottom of Figure 8, a row (34) is presented for bezafibrate (BZF), a known "right-shifting" allosteric hemoglobin modifier. As with all the newly discovered "right-shifting" allosteric hemoglobin modifiers, the hemoglobin treated with BZF had a higher P_{50} than the P_{50} for the control. Figure 8 shows the varying R_{2-6} moieties for the substituted phenyl compounds tested, and when a compound which did not have a substituted phenyl, the name of the compound is written across R_{2-6} (e.g., naphthyl, adamantyl, indanyl). The R_{7-8}

moieties were methyl groups for each compound tested and the R_9 moiety was a sodium cation for each compound tested (derived from the NaHCO_3 treatment prior to testing). Because other compounds within the family would have a similar mode of binding (e.g., those with different R_{2-9} moieties), their effect on the P_{50} value can be expected to be the same. The phthalimide structure defined by Figures 16a-b and Example 29 had a mean P_{50} value (e.g., $P_{50}\text{Drug}/P_{50}\text{Control}$) of 1.08 indicating the allosteric equilibrium for hemoglobin had been shifted towards favoring the low oxygen affinity "T" state of hemoglobin by the phthalimide compound.

Figure 9 shows the effect some of the compounds have on the oxygen dissociation of normal hemoglobin in intact human red blood cells (RBCs). The first entry provides the P_{50} value obtained for a control of human RBCs alone. The next two entries provide the P_{50} values when the RBCs are mixed together with a 10 millimolar (mM) solution of the sodium salt of either 2-[4-(((3,5-dichlorophenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($\text{C}_{18}\text{H}_{17}\text{Cl}_2\text{NO}_4$) (discussed in Example 10) or 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($\text{C}_{20}\text{H}_{23}\text{NO}_4$) (discussed in Example 15), respectively. Note that the P_{50} values for the hemoglobin in intact RBCs treated with the compounds is much greater than the P_{50} value for untreated hemoglobin under physiological conditions (e.g., the control of 27). In addition, it was determined that the P_{50} value was raised from 27 to 31 in the presence of 1 mM 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-

5 methyl propionic acid and to 42 in the presence of 2
mM 2-[4((((3,5-dimethylphenyl)amino)carbonyl)
methyl)phenoxy]-2-methyl propionic acid. This data
establishes the permeability of the of the compounds
to the cell membrane and that serum albumin does not
interfere with the drug's influence on the oxygen
dissociation curve of hemoglobin. Entries 23 and 24
in Figure 9 provide the P_{50} values for intact RBCs
treated with 10 mM of the same two compounds used in
10 entries 21 and 22, respectively, except that the
RBCs were washed with a 240 fold excess of 0.9%
saline. The relatively slight drop in the P_{50} value
after the saline wash, which represents a high
retention of allosteric effect, shows that the
15 compounds used in the present invention have high
binding affinity for hemoglobin.

Figure 10 is a graph illustrating the oxygen
dissociation curves produced when a 5.4 millimolar
solution of normal hemoglobin is tested at pH 7.4
20 using HEPES as the buffer in a Hem-O-Scan oxygen
dissociation analyzer. As described above, the P_{50}
values reported in Figure 8 were determined from
curves like those shown in Figure 10. With
particular reference to Figure 10, the percent
25 oxygen saturation (SO_2 on the vertical axis) is
plotted against the partial pressure of oxygen (PO_2
on the horizontal axis). Curve number 1 shows the
oxygen dissociation curve (ODC) in the absence of an
allosteric modifying agent. Curve number 2 shows
30 the ODC has been shifted to the right when 10 mM
bezafibrate (a known right shifting agent)
solubilized with an equimolar amount of $NaHCO_3$ is
added to the hemoglobin. It should be noted that as

the curve is right shifted to a lower oxygen affinity state, the P_{50} value increases. Curve number 3 shows the right shift caused by adding a 10 mM concentration of 2-[4-(((4-chlorophenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{18}H_{18}ClNO_4$) (described in Example 8 above) to the hemoglobin. Curve number 4 shows the right shift caused by adding a 10 mM concentration of 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{20}H_{23}NO_4$) (described in Example 15) to the hemoglobin. Finally, curve number 5 shows the right shift caused by adding a 10 mM concentration of 2-[4-(((3,5-dichlorophenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{18}H_{17}Cl_2NO_4$) (described in Example 10) to the hemoglobin. The right shifting effect shown in Figure 10 indicates the compounds may be used to lower the oxygen affinity of hemoglobin.

Figure 11 illustrates the effect of particular compounds on the ODC of whole human blood. Like Figure 10, the percent oxygen saturation is plotted against the partial pressure of oxygen. As described above, the P_{50} values reported in Figure 9 were determined from curves like those shown in Figure 11. For these curves, 50 μ l of whole human blood was mixed with a 50 μ l solution of the test compound in HEPES buffer at pH 7.4. Curve number 1 shows the ODC of hemoglobin in unreacted whole blood. Curves 2 and 3 respectively illustrate the right shifting effect of the salts of a 10 mM concentration of 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic

acid ($C_{20}H_{23}NO_4$) (described in Example 15) or a 10 mM concentration of 2-[4-(((3,5-dichlorophenyl)amino)carbonyl)methyl]phenoxy]-2-methyl propionic acid ($C_{18}H_{17}Cl_2NO_4$) (described in Example 10) on hemoglobin in whole blood.

Figure 12 shows ODC curves of human hemoglobin (5.4mM) in HEPES buffer at pH 7.4 which were made in a manner similar to that described in conjunction with Figure 10. Like Figures 10 and 11, the percent oxygen saturation is plotted against the partial pressure of oxygen. Curve number 1 shows ODC of human hemoglobin in the absence of any allosteric modifying agent. Curves 2 and 3 show the right shifting effect of 1 mM and 10 mM concentrations of 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl]phenoxy]-2-methyl propionic acid ($C_{20}H_{23}NO_4$) (described in Example 15) on human hemoglobin. Hence, this compound forces hemoglobin to a lower oxygen affinity state. Curve number 4 shows the right shifting effect of 2.5 mM 2,3-diphosphoglycerate (2,3-DPG), which is a natural allosteric hemoglobin effector. Curve number 5 shows the combined effect of two effectors, e.g., 1mM 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl]phenoxy]-2-methyl propionic acid and 2.5 mM 2,3-DPG, is greater than either effector alone. The synergistic effect may be utilized such that smaller quantities of drug are added to blood.

Figure 13 illustrates the utility of 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl]phenoxy]-2-methyl propionic acid (called RSR-13) in preserving the oxygen affinity of hemoglobin in stored blood. RSR-13, 1 mM and 2 mM, was added to

samples of human RBCs (packed cells) which were stored at 4°C in standard adsol formulation for 40-70 days. As can be seen from Figure 13, the ODC of untreated blood left-shifts over time (indicated by a drop in the P_{50} value) to a high oxygen affinity state. The increase in oxygen affinity of stored blood is attributed to a decreased concentration of 2,3-DPG. The P_{50} value of 40 day old untreated samples left shifted to 32; however, samples treated with 1mM RSR-13 remained relatively unchanged ($P_{50}=90$) and those treated with 2 mM RSR-13 were right shifted ($P_{50} = 45$). Figure 13 shows similar concentration dependent effects of RSR-13 on the ODCs of 50, 60, 70 day old packed cells. Because of the glycolytic metabolism, the pH of untreated red cells dropped over a period of time from 6.85 at 40 days to 6.6 for 70 day old samples and this would possibly explain the slight right shifting of untreated 70 day old samples compared to 40 day old samples under the Bohr effect. The pH of red blood cells treated with RSR-13 was consistently lower than untreated samples, which suggests that RSR-13 favorably decreases the rate of glycolytic metabolism. RSR-13 had no adverse effect on the stability of RBCs as evidenced by consistent RBC counts in treated and untreated samples. Similarly, the amount of hemolysis was consistent in both treated and untreated samples of packed cells.

Figure 14 shows the percentage oxygen delivered, ΔY , by packed cells. Changes in the oxygen saturation ΔY was calculated by Hill's equation (discussed in Stryer, *Biochemistry*, W.H. Freeman and Co., San Francisco, 1975, Chapter 4, pp,

71-94, which are herein incorporated by reference) at 100 to 30 torr. Column 1 shows the ΔY (59) corresponding to the untreated packed red blood cells. Column 2 shows the ΔY (50) of packed red blood cells stored for 40 days at 4°C in the best available adsol formulation. Column 3 shows that the $\Delta Y=58$ for 40 day old packed cells treated with RSR-13 (1 mM), which is comparable to fresh packed cells. Note, that the decrease (approximately 10%) in the oxygen delivery by packed cells is corrected by the addition of 1 mmol RSR-13.

Figure 15 shows the change in the P_{50} values of outdated packed red blood cells on treatment with 2-[4-(((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid (RSR-13). 50 μ l of 40, 50, and 60 day old red cells were mixed with 50 μ l of RSR-13 to give final concentrations of RSR-13 at 1 mmol and 2 mmol. Control samples were prepared by mixing 1:1 packed cells and buffer. As can be seen from Figure 15, the P_{50} value of untreated samples were consistently lower than samples treated with RSR-13. In addition, a comparison of the results for the fresh red cells with red cells which were aged 40, 50, and 60 days shows a sharp decline in P_{50} value with age. The P_{50} values of 40, 50, 60 day old red cell samples treated with 1 mmol RSR-13 were comparable to the $P_{50}=38$ value found for fresh red cells. These results show that the addition of RSR-13 to the stored red cells restores the cells oxygen affinity.

Since the compounds contemplated by this invention are capable of allosterically modifying hemoglobin so that a low oxygen affinity "T" state is

5 favored (right shifting the equilibrium curve as indicated by the P_{50} column in Figures 8-9), these compounds will be useful in treating a variety of disease states in mammals including humans where tissues suffer from low oxygen tension, such as cancer and ischemia. As pointed out by Hirst et al. in Radiat. Res., Vol. 112, (1987), pp. 164, decreasing the oxygen affinity of hemoglobin in circulating blood has been shown to be beneficial in the radiotherapy of tumors. The compounds may be administered to patients in whom the affinity of hemoglobin for oxygen is abnormally high. Particular conditions include certain hemoglobinopathies and certain respiratory distress syndromes in new born infants aggravated by high fetal hemoglobin levels and when the availability of hemoglobin/oxygen to the tissues is decreased (e.g., in ischemic conditions such as peripheral vascular disease, coronary occlusion, cerebral vascular accidents, or tissue transplant). The compounds may also be used to inhibit platelet aggregation and may be used for antithrombotic purposes and wound healing. Topical application could be used for wound healing. In addition, the compounds may be used to treat low oxygen related disorders in the brain such as Alzheimer's disease, depression, and schizophrenia. It may be desirable to administer the compounds to a patient prior to and/or simultaneously with the transfusion of the treated whole blood or red blood cells in order to avoid substantial variations in the hemoglobin oxygen affinity due to dilution that occurs when the blood is administered.

The compounds can be added to whole blood or packed cells preferably at the time of storage or at the time of transfusion in order to facilitate the dissociation of oxygen from hemoglobin and improve the oxygen delivering capability of the blood. Preferably, the compounds would be added in an amount of about 50 mg to 1 g per unit of blood (473 ml) or unit of packed cells (235 ml). When blood is stored, the hemoglobin in the blood tends to increase its affinity for oxygen by losing 2,3-diphosphoglycerides. As described above, the compounds of this invention are capable of reversing and/or preventing the functional abnormality of hemoglobin which is observed when whole blood or packed cells are stored. The compounds may be added to whole blood or red blood cell fractions in a closed system using an appropriate reservoir in which the compound is placed prior to storage or which is present in the anticoagulating solution in the blood collecting bag.

Administration can be achieved orally, by intravenous or intraperitoneal injection, or rectally by suppository where the dose and the dosing regiment is varied according to individual sensitivity and the type of disease state being treated. Studies with mice have shown that a mg/kg/day dose of 2-[4((((3,5-dimethylphenyl)amino)carbonyl)methyl)phenoxy]-2-methyl propionic acid ($C_{20}H_{23}NO_4$) (discussed in Example 15) given intraperitoneally is well tolerated. If the compounds are used for wound healing, the compounds could advantageously be applied topically directly to the wound area. In addition, the compounds can

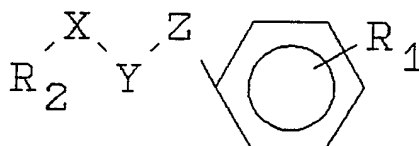
be mixed with blood external to a patient's body prior to and/or simultaneously with a transfusion. The compounds can be administered in the pure form or in a pharmaceutically acceptable formulation including suitable elixirs, binders, and the like or as pharmaceutically acceptable salts or other derivatives. It should be understood that the pharmaceutically acceptable formulations and salts include liquid and solid materials conventionally utilized to prepare injectable dosage forms and solid dosage forms such as tablets and capsules. Water may be used for the preparation of injectable compositions which may also include conventional buffers and agents to render the injectable composition isotonic. Solid diluents and excipients include lactose starch, conventional disintegrating agents, coatings and the like.

While the invention has been described in terms of its preferred embodiments, those skilled in the art will recognize that the invention can be practiced with modification within the spirit and scope of the appended claims.

CLAIMS

Having thus described our invention, what we claim as new and desire to secure by Letters Patent is as follows:

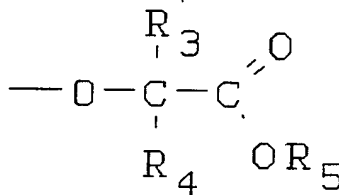
- 5 1. A method for allosterically modifying hemoglobin comprising the step of exposing hemoglobin to a compound of the general structural formula:



- 10 where R_2 is a substituted or unsubstituted aromatic compound, or a substituted or unsubstituted alkyl ring compound, or a substituted or unsubstituted phthalimide compound where X is a carboxyl, Y is a nitrogen and R_2 completes the phthalimide compound by being bonded to both X and Y,

- 15 and where X, Y, and Z are CH_2 , NH, CO, O or N with the caveat that the X, Y, and Z moieties are each different from one another,

and where R_1 has the formula:

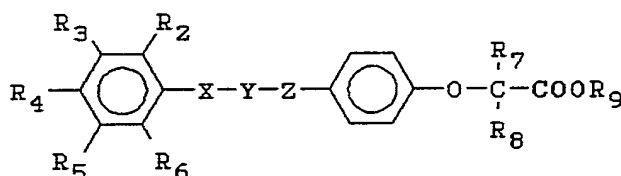


where R_1 can be connected to any position on the phenyl ring, and

where R_3 and R_4 are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R_3 and R_4 , and

where R_5 is a hydrogen, halogen, C_{1-3} loweralkyl, or a salt cation.

2. A method for allosterically modifying hemoglobin comprising the step of exposing hemoglobin to a compound of the general structural formula:



wherein X , Y and Z may each be CH_2 , CO , NH or O , with the caveat that the X , Y , and Z moieties are each different from one another,

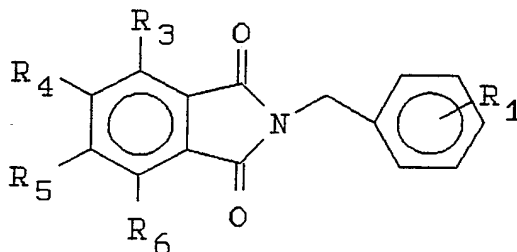
and wherein R_{2-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{2-6} sites,

and wherein R_{7-8} are hydrogen, methyl or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R_7 and R_8 ,

and wherein R_9 is a hydrogen, halogen, substituted or unsubstituted C_{1-3} loweralkyl, or a

salt cation.

3. A method for allosterically modifying hemoglobin comprising the step of exposing hemoglobin to a compound of the general structural formula:

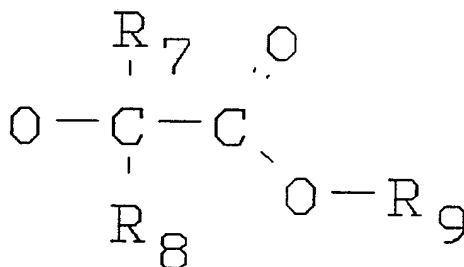


5

wherein R_{3-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{3-6} sites, and

10

wherein R_1 has the formula:



where R_1 can be connected to any position on the phenyl ring, and

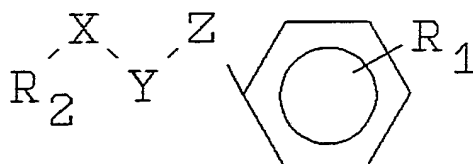
15

where R_7 and R_8 are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same

or different, or alkyl moieties as part of an aliphatic ring connecting R_7 and R_8 , and

where R_9 is a hydrogen, halogen, C_{1-3} loweralkyl, or a salt cation.

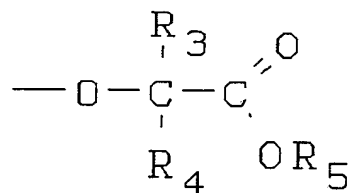
- 5 4. A method for treating blood such that hemoglobin in said blood is allosterically modified towards a low oxygen affinity state, comprising the step of exposing said blood to a compound of the general structural formula:



- 10 where R_2 is a substituted or unsubstituted aromatic compound, or a substituted or unsubstituted alkyl ring compound, or a substituted or unsubstituted phthalimide compound where X is a carboxyl, Y is a nitrogen and R_2 completes the phthalimide compound by being bonded to both X and Y ,

15 and where X , Y , and Z are CH_2 , NH , CO , O or N with the caveat that the X , Y , and Z moieties are each different from one another,

- 20 and where R_1 has the formula:

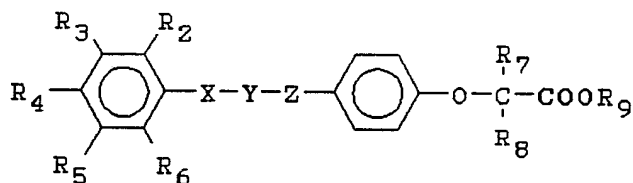


where R_1 can be connected to any position on the phenyl ring, and

where R_3 and R_4 are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R_3 and R_4 , and

where R_5 is a hydrogen, halogen, C_{1-3} loweralkyl, or a salt cation.

5. A method for treating blood such that hemoglobin in said blood is allosterically modified towards a low oxygen affinity state, comprising the step of exposing said blood to a compound of the general structural formula:



wherein X, Y and Z may each be CH_2 , CO, NH or O, with the caveat that the X, Y, and Z moieties are each different from one another, and wherein R_{2-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be

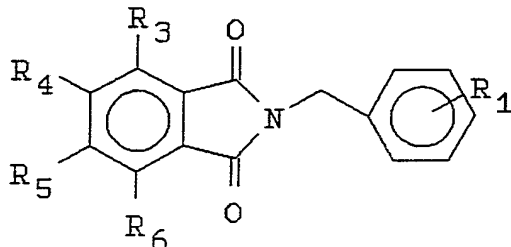
the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{2-6} sites,

5 and wherein R_{7-8} are hydrogen, methyl or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R_7 and R_8 ,

10 and wherein R_9 is a hydrogen, halogen, substituted or unsubstituted C_{1-3} loweralkyl, or a salt cation.

6. A method for treating blood such that hemoglobin in said blood is allosterically modified towards a low oxygen affinity state, comprising the step of exposing said blood to a compound of the general structural formula:

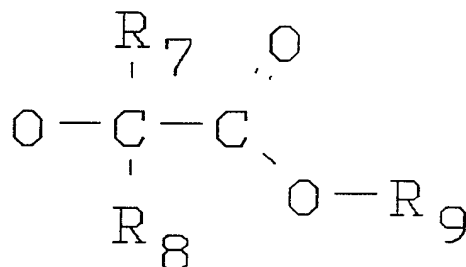
15



wherein R_{3-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{3-6} sites, and

20

wherein R_1 has the formula:

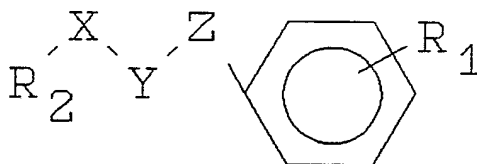


where R_1 can be connected to any position on the phenyl ring, and

where R_7 and R_8 are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an

where R_9 is a hydrogen, halogen, C_{1-3} loweralkyl, or a salt cation.

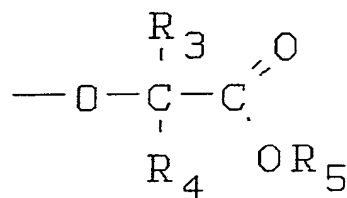
7. A method of storing blood, comprising the steps of exposing blood to be stored to a compound of the general structural formula:



where R_2 is a substituted or unsubstituted aromatic compound, or a substituted or unsubstituted alkyl ring compound, or a substituted or unsubstituted phthalimide compound where X is a carboxyl, Y is a nitrogen and R_2 completes the

phthalimide compound by being bonded to both X and Y,

and where X, Y, and Z are CH₂, NH, CO, O or N with the caveat that the X, Y, and Z moieties are each different from one another,
and where R₁ has the formula:

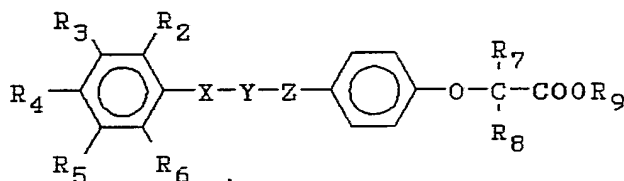


where R₁ can be connected to any position on the phenyl ring, and

where R₃ and R₄ are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R₃ and R₄, and

where R₅ is a hydrogen, halogen, C₁₋₃ loweralkyl, or a salt cation.

8. A method of storing blood, comprising the steps of exposing blood to be stored to a compound of the general structural formula:



wherein X, Y and Z may each be CH₂, CO, NH or O, with the caveat that the X, Y, and Z moieties are

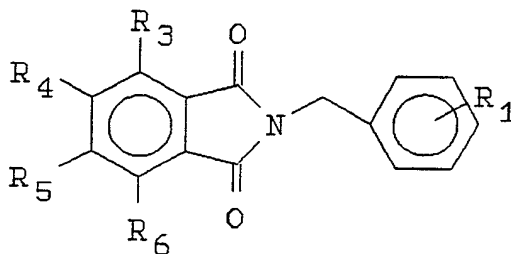
each different from one another,

and wherein R_{2-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{2-6} sites,

and wherein R_{7-8} are hydrogen, methyl or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R_7 and R_8 ,

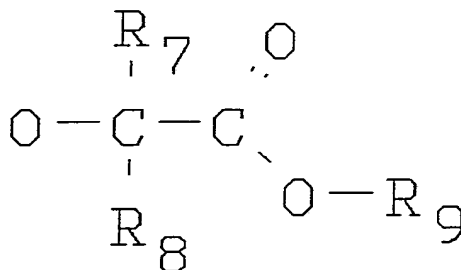
and wherein R_9 is a hydrogen, halogen, substituted or unsubstituted C_{1-3} loweralkyl, or a salt cation.

9. A method of storing blood, comprising the steps of exposing blood to be stored to a compound of the general structural formula:



wherein R_{3-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{3-6} sites, and

wherein R_1 has the formula:

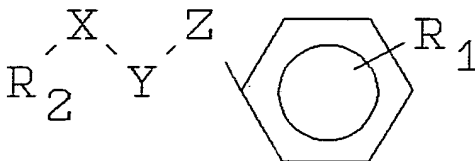


where R_1 can be connected to any position on the phenyl ring, and

where R_7 and R_8 are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an

where R_9 is a hydrogen, halogen, C_{1-3} loweralkyl, or a salt cation.

10. A method of restoring the oxygen affinity of red blood cells, comprising the steps of
- storing red blood cells for a period of time;
- and
- exposing said red blood cells to a compound of the general structural formula:

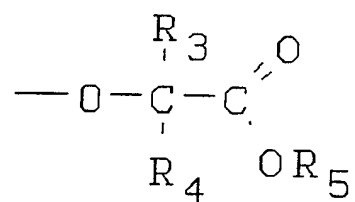


where R_2 is a substituted or unsubstituted aromatic compound, or a substituted or unsubstituted

alkyl ring compound, or a substituted or unsubstituted phthalimide compound where X is a carboxyl, Y is a nitrogen and R₂ completes the phthalimide compound by being bonded to both X and Y,

and where X, Y, and Z are CH₂, NH, CO, O or N with the caveat that the X, Y, and Z moieties are each different from one another,

and where R₁ has the formula:



where R₁ can be connected to any position on the phenyl ring, and

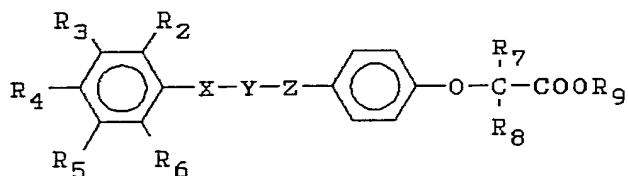
where R₃ and R₄ are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R₃ and R₄, and

where R₅ is a hydrogen, halogen, C₁₋₃ loweralkyl, or a salt cation.

11. A method of restoring the oxygen affinity of red blood cells, comprising the steps of

storing red blood cells for a period of time; and

exposing said red blood cells to a compound of the general structural formula:



wherein X, Y and Z may each be CH₂, CO, NH or O, with the caveat that the X, Y, and Z moieties are each different from one another,

and wherein R₂₋₆ are either hydrogen, halogen,
 5 or a substituted or unsubstituted C₁₋₃ alkyl group,
 or a C₁₋₃ ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R₂₋₆ sites,

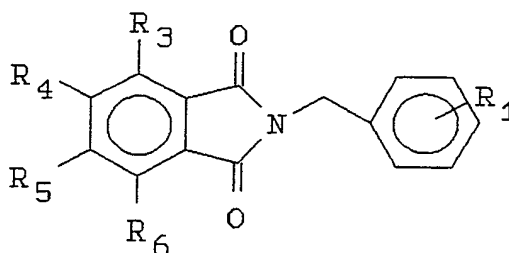
10 and wherein R₇₋₈ are hydrogen, methyl or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R₇ and R₈,

and wherein R₉ is a hydrogen, halogen,
 15 substituted or unsubstituted C₁₋₃ loweralkyl, or a salt cation.

12. A method of restoring the oxygen affinity of red blood cells, comprising the steps of

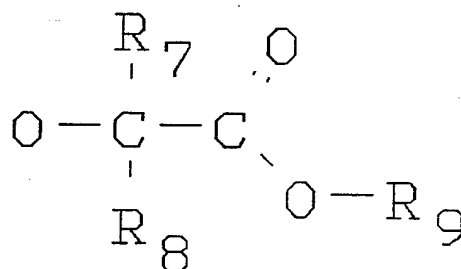
20 storing red blood cells for a period of time;
 and

exposing said red blood cells to a compound of the general structural formula:



wherein R_{3-6} are either hydrogen, halogen, or a substituted or unsubstituted C_{1-3} alkyl group, or a C_{1-3} ether or ester, and these moieties may be the same or different, or alkyl moieties of an aromatic or aliphatic ring incorporating two of the R_{3-6} sites, and

wherein R_1 has the formula:



where R_1 can be connected to any position on the phenyl ring, and

where R_7 and R_8 are hydrogen, halogen, methyl, or ethyl groups and these moieties may be the same or different, or alkyl moieties as part of an aliphatic ring connecting R_7 and R_8 , and

where R_9 is a hydrogen, halogen, C_{1-3} loweralkyl, or a salt cation.

FIG. 1a

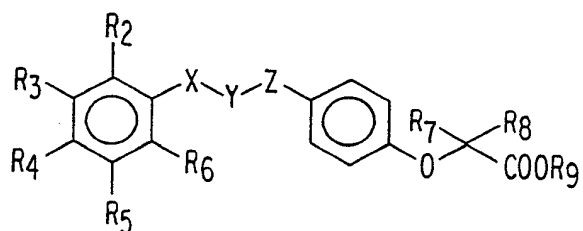


FIG. 1b

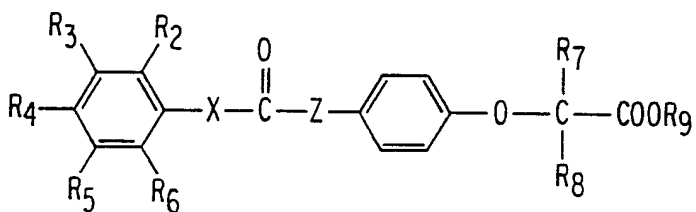


FIG. 1c

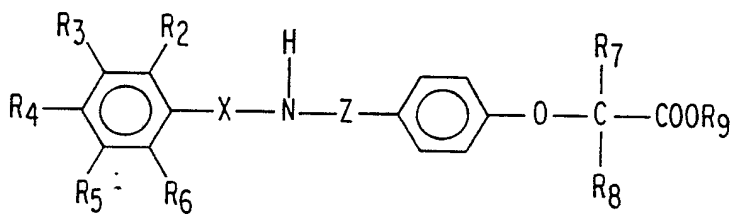


FIG. 2A

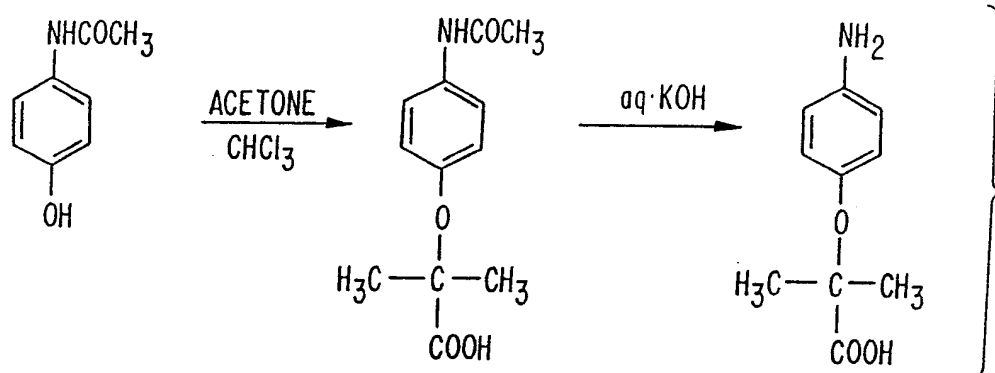


FIG. 2B

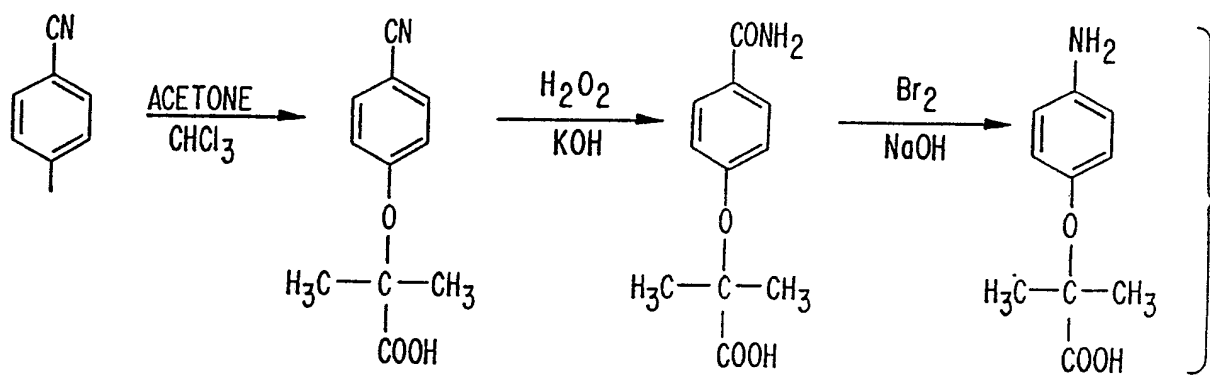


FIG. 2C

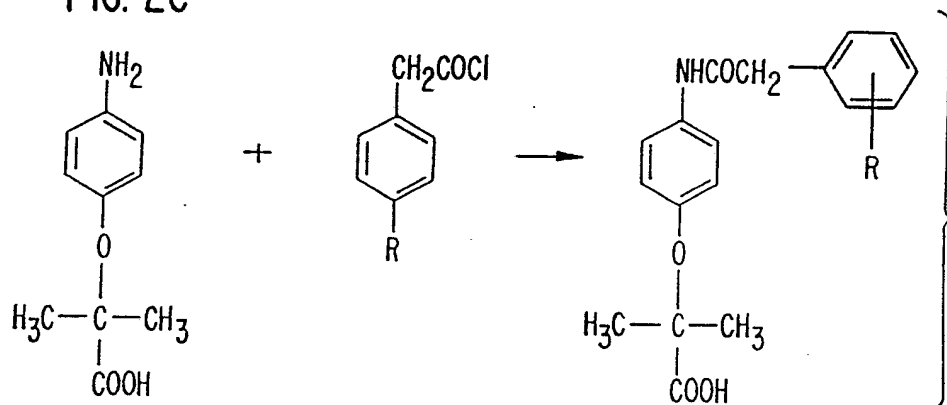


FIG. 3

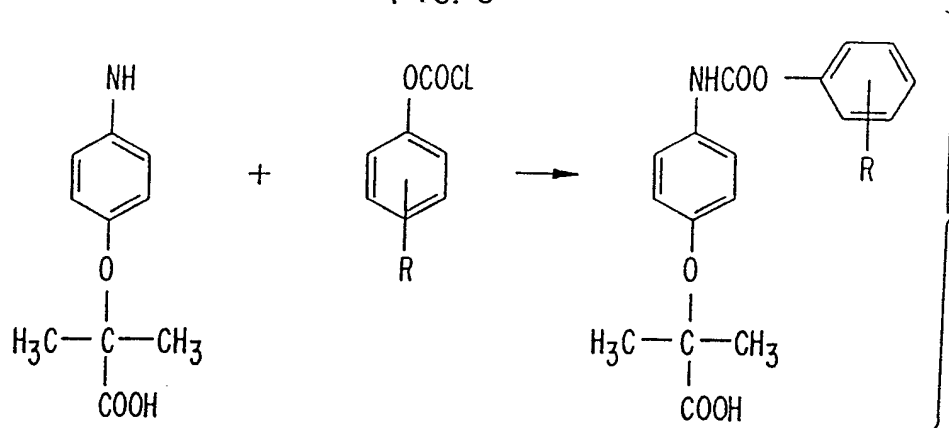


FIG. 4

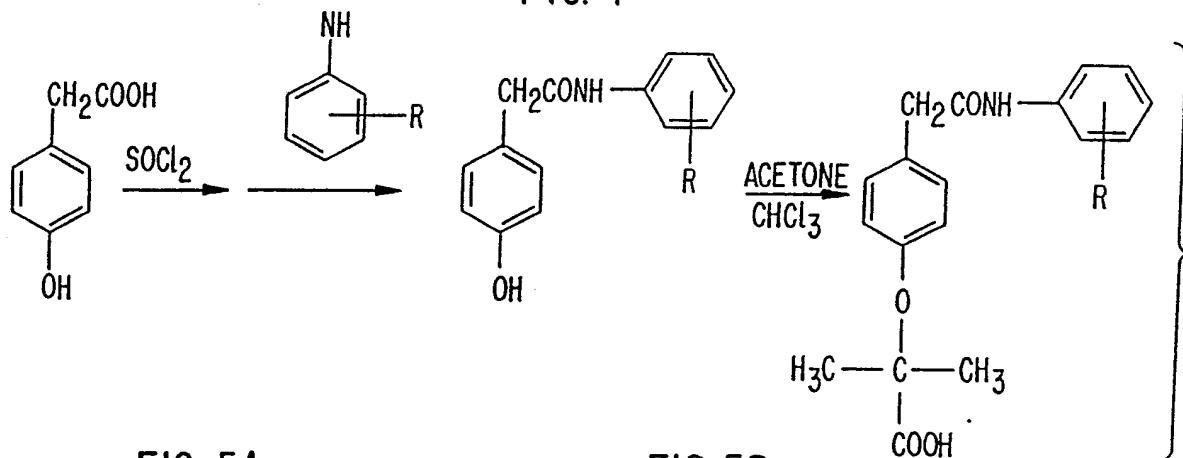


FIG. 5A

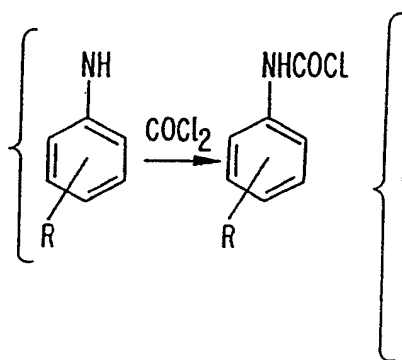


FIG. 5B

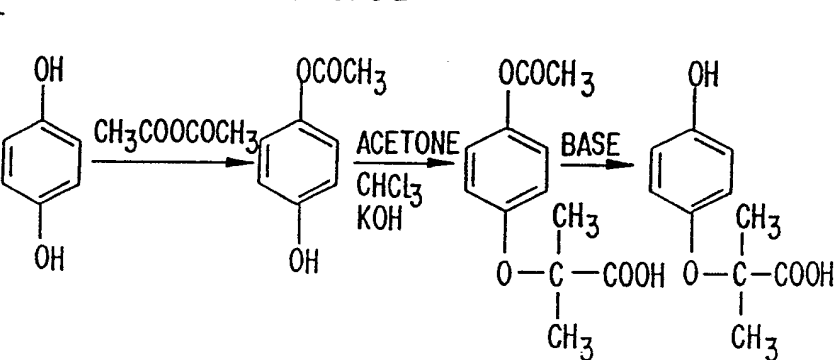
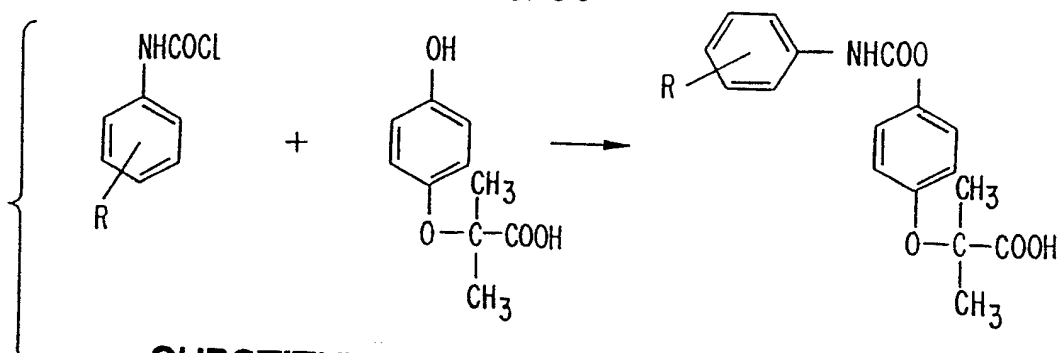


FIG. 5C



SUBSTITUTE SHEET

FIG. 6a

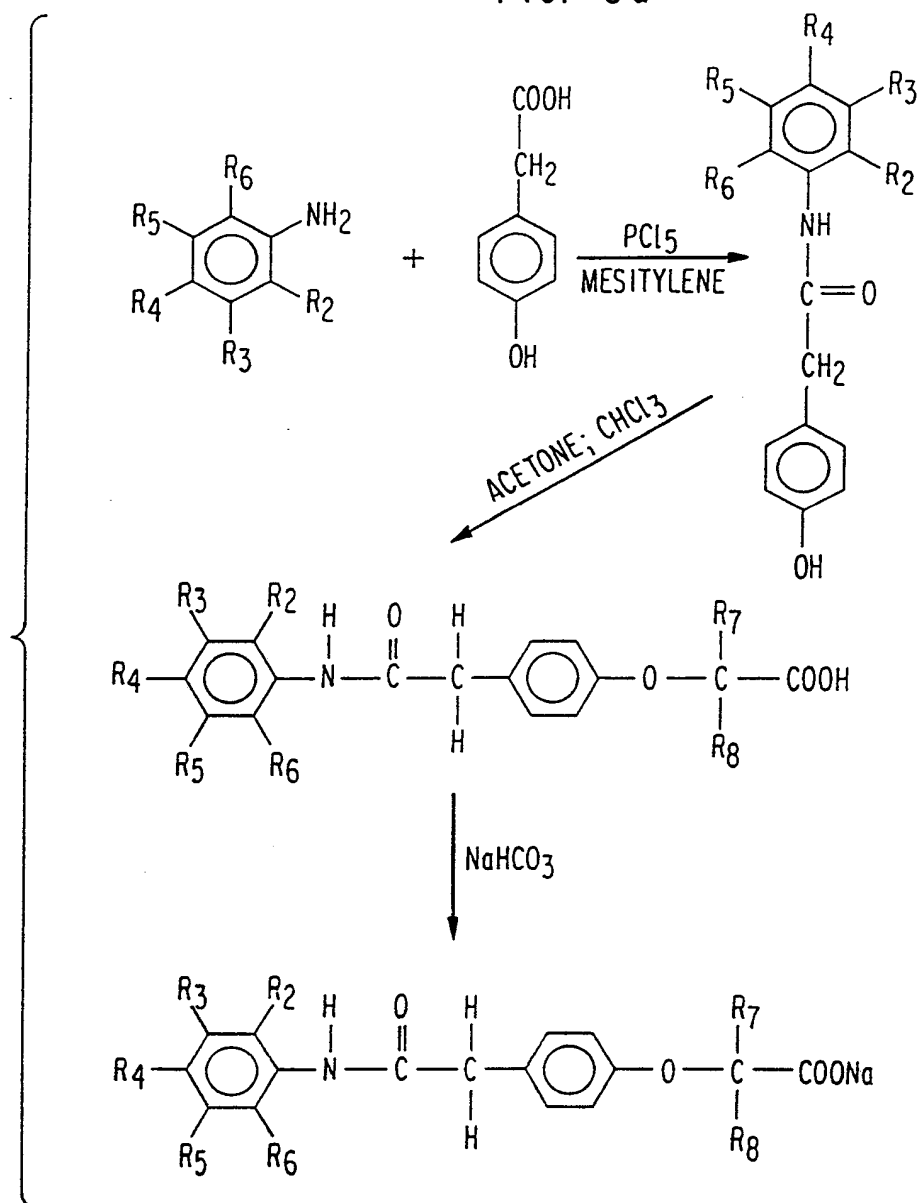


FIG. 7a

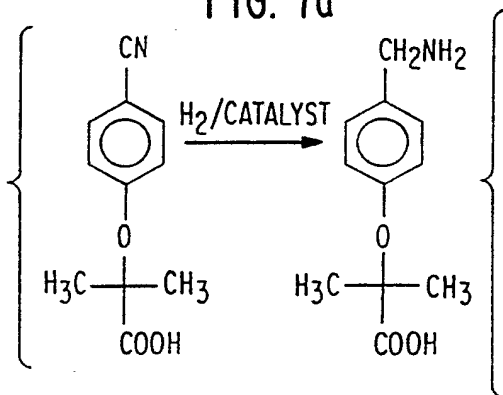


FIG. 7b

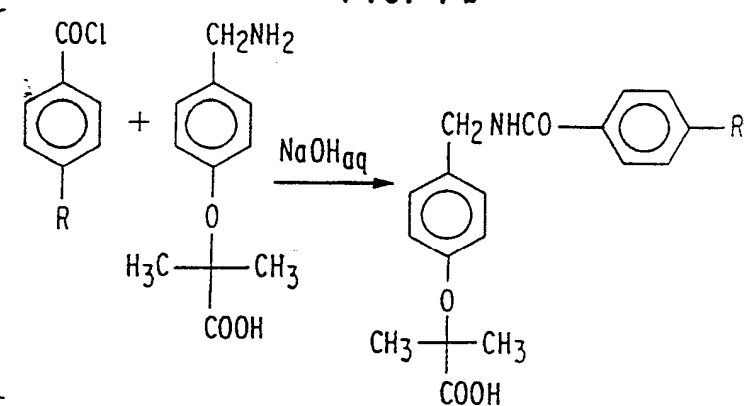


FIG. 6b

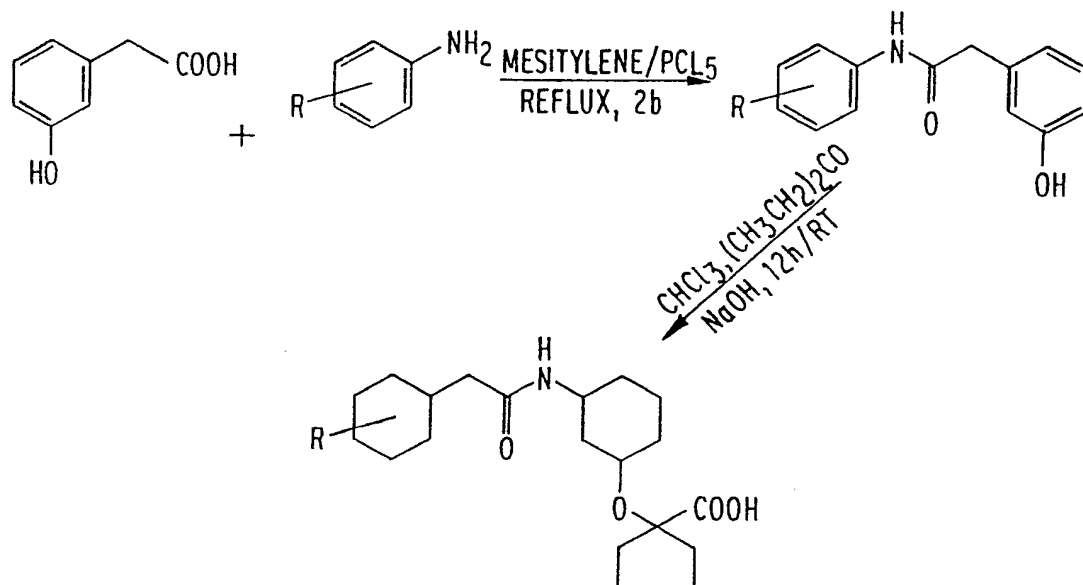


FIG. 16a

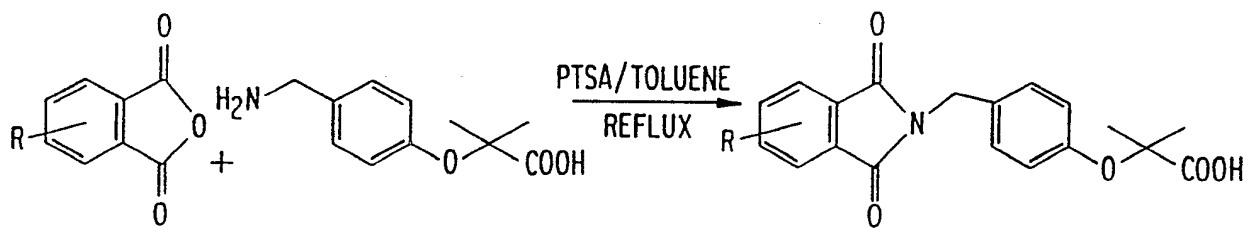


FIG. 16b

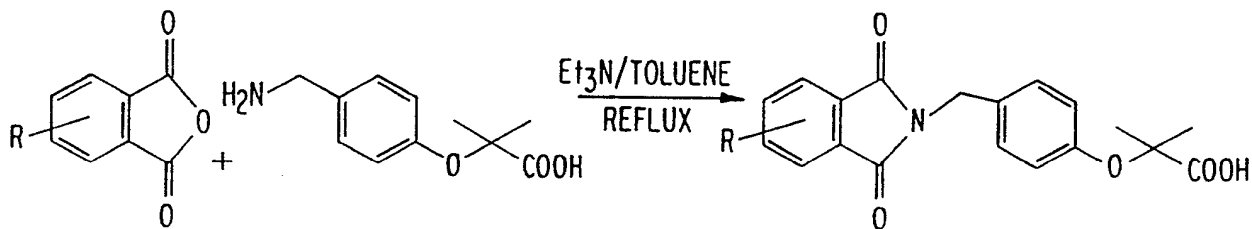


FIG. 8

COMP. No.	R ₂	R ₃	R ₄	R ₅	R ₆	X	Y	Z	P ₅₀ (CONTROL)	P ₅₀	P ₅₀ /P _{50C}
1	H	H	H	H	H	CO	NH	CH ₂	18	35	1.94
2	CL	H	H	H	H	CO	NH	CH ₂	18	27.5	1.52
3	H	CL	H	H	H	CO	NH	CH ₂	18	37.5	2.08
4	H	H	CL	H	H	CO	NH	CH ₂	19	48	2.52
5	H	CL	CL	H	H	CO	NH	CH ₂	18	40.5	2.25
6	H	CL	H	CL	H	CO	NH	CH ₂	18	47	2.60
7	H	CL	CL	CL	H	CO	NH	CH ₂	19	40	2.10
8	H	H	H	H	H	CH ₂	CO	NH	19	35	1.73
9	H	CL	H	H	H	CH ₂	CO	NH	18	44	2.44
10	H	CL	H	H	H	CH ₂	CO	NH	18	44	2.31
11	H	H	F	H	H	CH ₂	CO	NH	18	35	1.94
12	H	H	CH ₃	H	H	CH ₂	CO	NH	18	45	2.50
13	H	H	CH ₃	H	H	CH ₂	CO	NH	18	42	2.33
14	H	H	OME	H	H	CH ₂	CO	NH	18	38	2.11
15	H	CL	CL	H	H	CH ₂	CO	NH	18	50	2.77
15	H	ME	H	ME	H	CH ₂	CO	NH	18	52	2.88
16	H	H	H	H	H	O	CO	NH	18	26.5	1.47
17	H	H	CL	H	H	O	CO	NH	19	34	1.78
18	H	H	H	H	H	NH	CO	CH ₂	19	54	2.84
19	H	H	CL	H	H	NH	CO	CH ₂	19	54	2.84
20	H	CL	CL	H	H	NH	CO	CH ₂	18	65	3.61
21	H	CL	H	CL	H	NH	CO	CH ₂	19	83	4.36
22	H	CL	CL	CL	H	NH	CO	CH ₂	19	63	3.30
23	H	H	F	H	H	NH	CO	CH ₂	18	45	2.50
24	H	H	CF ₃	H	H	NH	CO	CH ₂	18	44	2.44
25	H	H	CH ₃	H	H	NH	CO	CH ₂	18	49	2.72
26	H	CH ₃	H	CH ₃	H	NH	CO	CH ₂	19	75	3.94
27	CL	H	H	H	CL	NH	CO	CH ₂	18	34	1.89
28	H	ME	ME	H	H	NH	CO	CH ₂	18	62	3.41
29	-	-NAPHTHYL-	-	-	-	NH	CO	CH ₂	18	58	3.20
30	H	-PROPYL-	-	H	H	NH	CO	CH ₂	18	64	3.56
27	H	CL	H	H	H	NH	CO	CH ₂	18	61	3.40
28	H	H	CL	H	H	CH ₂	NH	CO	18	27	1.50
29	H	H	CH ₃	H	H	CH ₂	NH	CO	19	28	1.47
30	H	CL	CL	H	H	CH ₂	NH	CO	18	28	1.56
31	H	H	H	H	H	CH ₂	NH	CO	19	22	1.16
32	-	-INDANYL-	-	-	-	NH	CO	CH ₂	18	64	3.56
33	-	-ADAMANTYL-	-	-	-	NH	CO	CH ₂	18	32	1.78
34 BZF	H	H	CL	H	H	CO	NH	(CH ₂) ₂	18	33	1.83

SUBSTITUTE SHEET

FIG. 9

NO.	R ₂	R ₃	R ₄	R ₅	R ₆	x	z	P ₅₀
CONTROL	-	-	-	-	-	-	-	27
21	H	CH ₃	H	CH ₃	H	NH	CH ₂	83
22	H	Cl	H	Cl	H	NH	CH ₂	87
23	H	CH ₃	H	CH ₃	H	NH	CH ₂	76
24	H	CH ₃	H	CH ₃	H	NH	CH ₂	68

FIG. 10

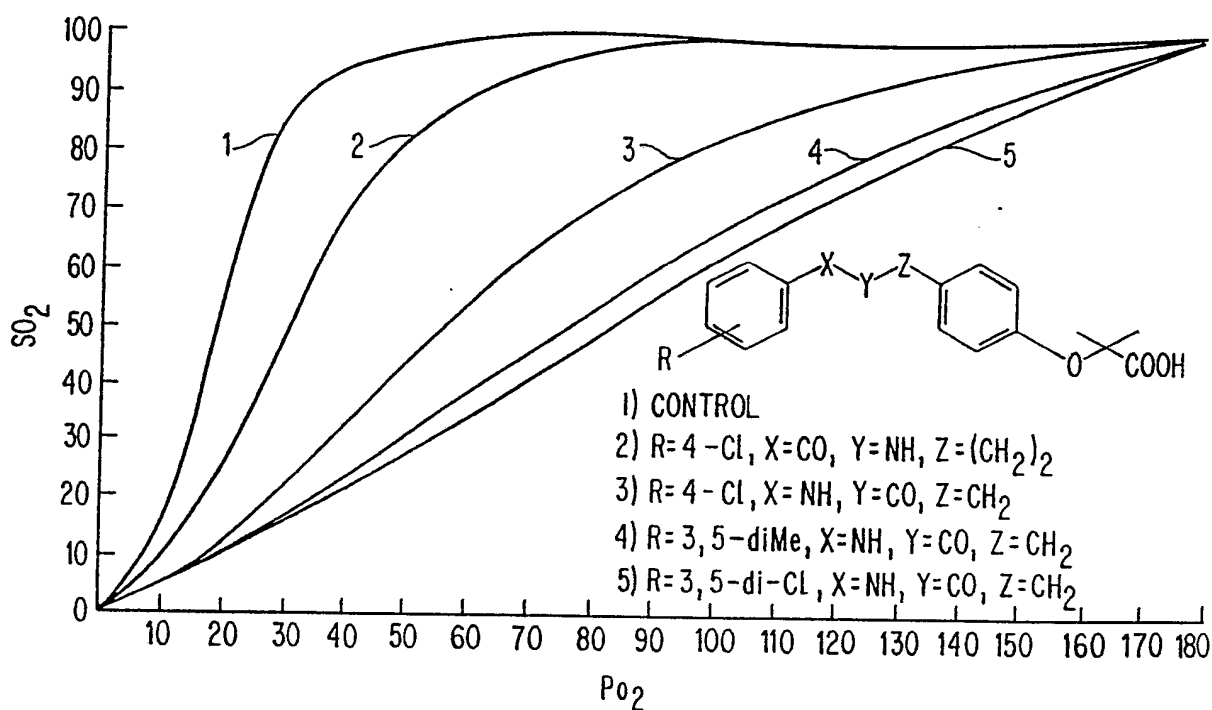


FIG. 11

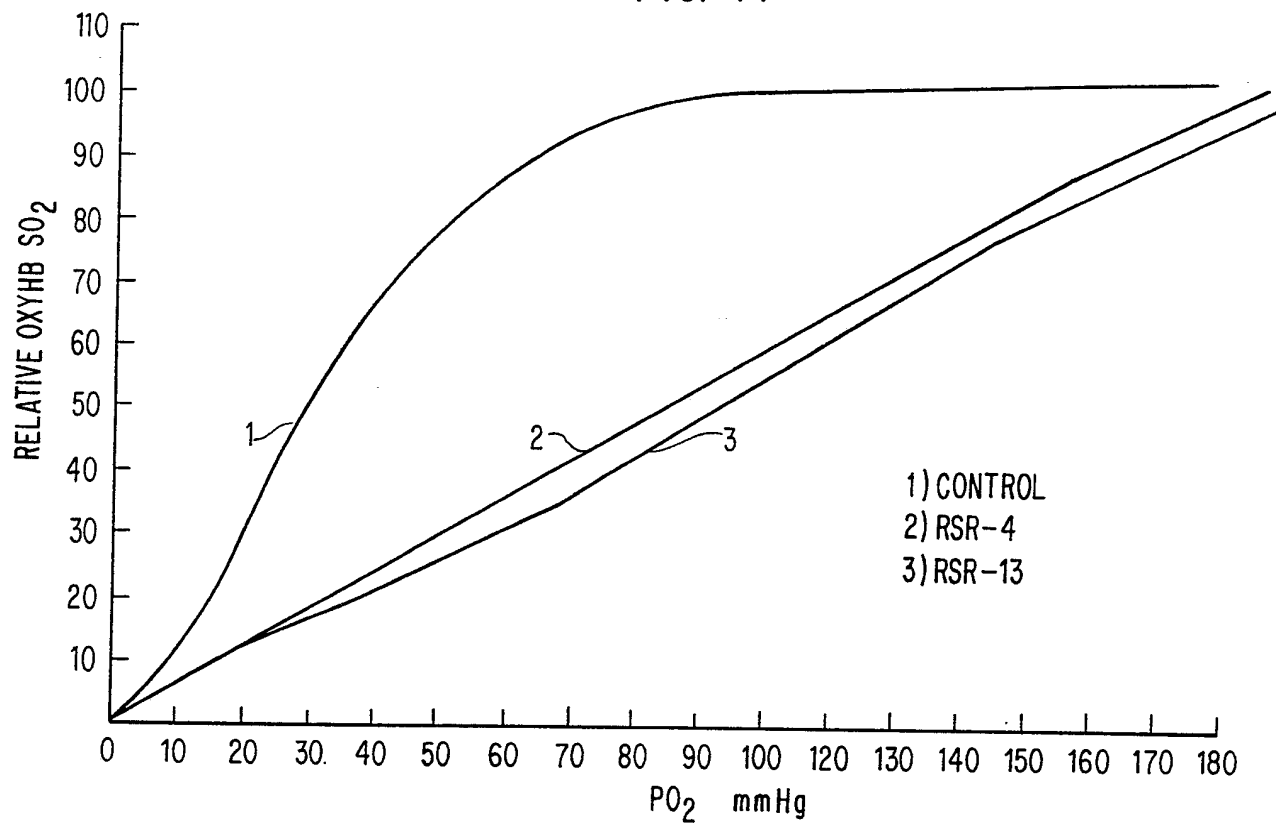


FIG. 12

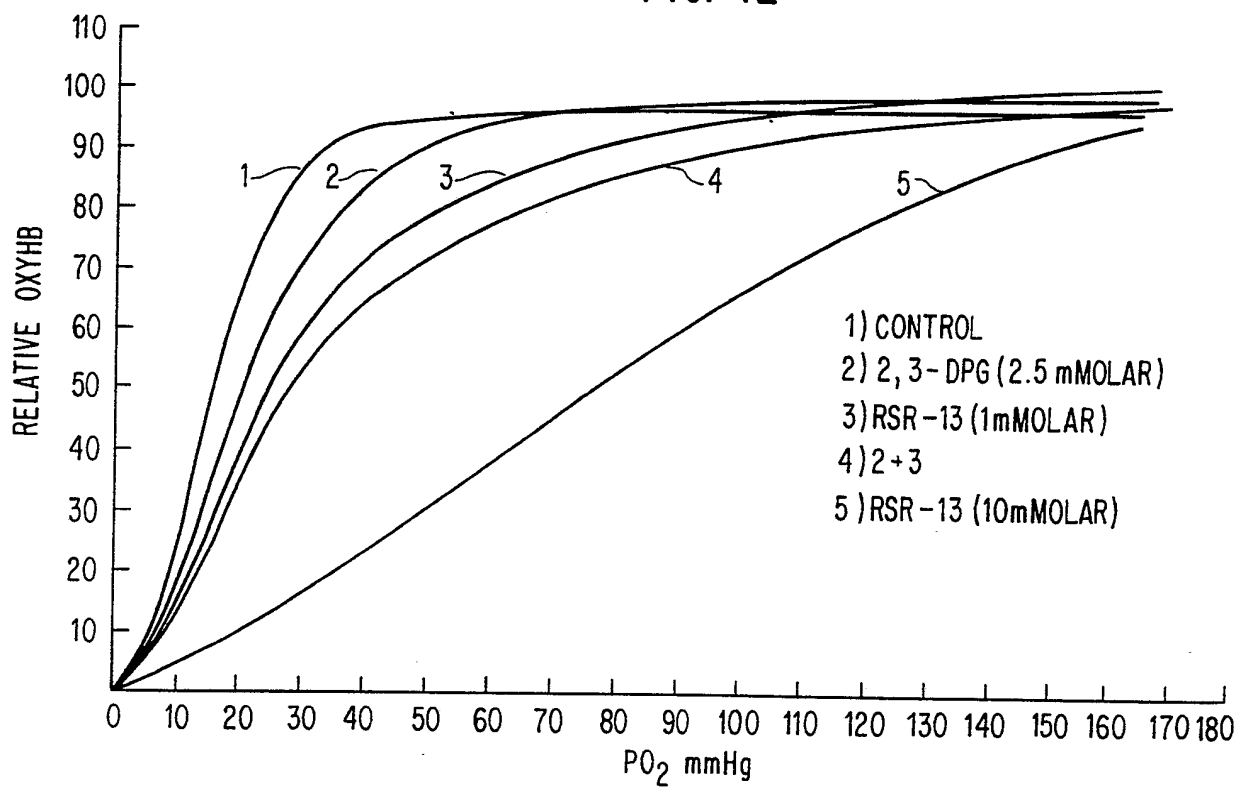


FIG. 13

PACKED RBC DAY OLD	P ₅₀ IN PRESENCE OF RSR-13		
	0mM	1mM	2mM
FRESH	38	—	—
40	32	39	45
50	33	39	45
60	34	40	47
70	35	39	50

FIG. 14

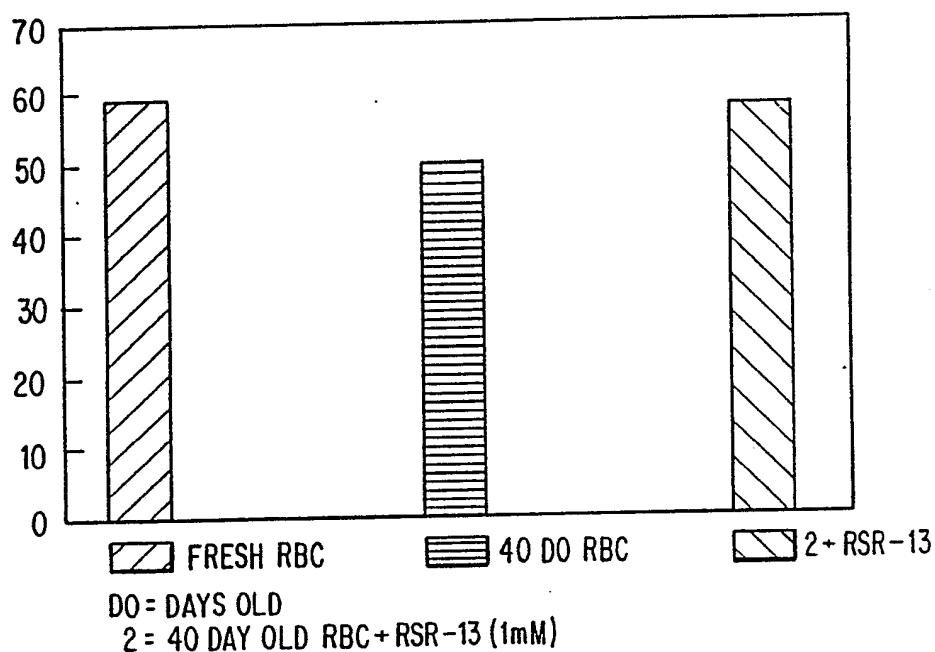


FIG. 15

PACKED RBC DAY OLD	P ₅₀ UPON ADDITION OF RSR-13		
	0mM	1mM	2mM
FRESH	38	—	—
40	32	38	42
50	31	38	45
60	34	39	46

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US92/04229

A. CLASSIFICATION OF SUBJECT MATTER

IPC(5) : A61K 31/245; A61K 31/195; A61K 31/325; C07C 45/00

US CL : Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 548/473,480

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

REGISTRY DATA BASE: SEARCH STRUCTURAL FORMULA OF CLAIM 3.

CAS DATA BASE: SEARCH TERM "ALLOSTERIC HEMOGLOBIN MODIFIERS"

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US, A, 4,704,402 (ABRAHAM ET AL.) 03 November 1987, See column 3, lines 52-68 through column 4, lines 1-65.	1-7 and 10-12
P,X	US, A, 5,049,695 (ABRAHAM ET AL.) 17 September 1991, See entire document.	1,2,4,5, 7,8,10 & 11
T	US, A, 5,122,539 (ABRAHAM ET AL.) 16 June 1992, See entire document.	1-2,4-5, 7-8 & 10-11
A	US, A, 4,699,926 (ABRAHAM ET AL.) 13 October 1987, See Example 28.	1-7 & 10-12



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents:

A document defining the general state of the art which is not considered to be part of particular relevance

E earlier document published on or after the international filing date

L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

O document referring to an oral disclosure, use, exhibition or other means

P document published prior to the international filing date but later than the priority date claimed

T

later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X

document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y

document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

&

document member of the same patent family

Date of the actual completion of the international search

24 AUGUST 1992

Date of mailing of the international search report

15 OCT 1992

Name and mailing address of the ISA/
Commissioner of Patents and Trademarks
Box PCT
Washington, D.C. 20231

Authorized officer

P. NAZARIO-GONZALEZ

Facsimile No. NOT APPLICABLE

Telephone No. (703) 308-1235

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US92/04229

A. CLASSIFICATION OF SUBJECT MATTER:
US CL :

514/512,513,533,535,538,833,486,488,490,563; 560/30,31,32; 568/452,455